

Academic Chemistry 2nd Semester Final 2022

Thermodynamics

13. The average kinetic energy of the particles in solids, liquids, and gases is measured as?

- a. Temperature
- b. Density
- c. Volume
- d. Pressure

Use the graph below to answer questions 14-16.



14. What phase change is occurring at 4?

- a. melting
- b. freezing
- c. vaporization
- d. none of the above

15. Is this diagram representing an exothermic or endothermic reaction?

- a. exothermic
- b. endothermic

16. Which stage of the graph would you find water as both a liquid and a solid?

- a. 2
- b. 3
- c. 4
- d. 5

17. During a phase change, the temperature:

- a. increases
- b. stays the same
- c. decreases

18. A sample of gold absorbs 50.6 Joules, causing its temperature to rise from 25.0°C to 82.7°C. If the specific heat capacity of gold is 0.130 J/g°C, what is the mass of the sample?

- a. 0.148 grams
- b. 8.16 grams
- c. 6.75 grams
- d. 22.2 grams

Chemistry 2nd Semester Finals

L Towner



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