

Oefeningen leereenheid : Redoxevenwichten.

De chemische cel – de sterkte van een reductor en oxidator – verloop van een reactie voorspellen – corrosie – elektrolyse

1. Men maakt in standaardvoorwaarden (conc. van stoffen is 1.0 M, temperatuur en druk bij normale omstandigheden) een **chemische cel met de metalen magnesium in halfcel 1 en het metaal lood in halfcel 2. De metaalelektroden zijn ondergedompeld in oplossingen van respectievelijk magnesiumnitraat en lood(II)nitraat. Vul volgende zinnen aan:**

Tussen de metalen en hun oplossing stellen zich de volgende evenwichten in:
halfcel 1: $\text{Mg} \rightleftharpoons \text{Mg}^{2+} + 2 \text{e}^-$ en halfcel 2: $\text{Pb} \rightleftharpoons \text{Pb}^{2+} + 2 \text{e}^-$

Aangezien het metaal **magnesium** de sterkste reductor is, is in die halfcel het evenwicht meer naar **rechts** verschoven. In de chemische cel gedraagt het metaal **lood** zich als pluspool en het metaal **magnesium** zich als minpool van de spanningsbron.

Bij stroomlevering in deze chemische cel lopen:

- in de metaalbrug **elektronen** van **Mg** naar **Pb**.

In het algemeen mogen we zeggen : ze lopen van halfcel met de

lagere E° -waarde naar de halfcel die de oxidator bevat met de **hogere** E° -waarde.

- **Positieve** ionen lopen in de elektrolytbrug van halfcel 1 naar halfcel 2,
negatieve ionen van halfcel 2 naar halfcel 1.

De reactievergelijkingen van de chemische processen in beide halfcellen zijn:
halfcel 1: $\text{Mg} \rightleftharpoons \text{Mg}^{2+} + 2 \text{e}^-$ en halfcel 2: $\text{Pb}^{2+} + 2 \text{e}^- \rightleftharpoons \text{Pb}$

De gebruikelijke verkorte notatie van het galvanisch element is : **Mg/Mg²⁺//Pb²⁺/Pb**

Hierbij komt de dubbele streep overeen met de **electrolytbrug/zoutbrug**. De bronspanning in deze chemische cel is gelijk aan **2,25 V**.

Compendium Van De Psychologie Deel 22 Motoriek

S Ashworth



Compendium Van De Psychologie Deel 22 Motoriek:

Boekblad ,1988

Reviewing **Compendium Van De Psychologie Deel 22 Motoriek**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Compendium Van De Psychologie Deel 22 Motoriek**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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