

## Chapter 2 The Nature of Matter • Section 1 Summary

## Describing Matter

### Key Concepts

- What kinds of properties are used to describe matter?
- What are elements, and how do they relate to compounds?
- What are the properties of a mixture?

**Matter** is anything that has mass and takes up space. **Chemistry** is the study of the properties of matter and how matter changes. In chemistry, a **substance** is a single kind of matter that is pure.

Every form of matter has two kinds of properties—**physical properties** and **chemical properties**. A **physical property** is observed without changing a substance into another substance. Examples of physical properties are hardness, texture, color, and ability to dissolve in water. A **chemical property** is the ability of a substance to change into different substances. Some chemical properties are burning and rusting.

All matter is made up of elements. An **element** is a pure substance that cannot be broken down into any other substance. **Elements are the simplest substances**. Each element is identified by its specific physical and chemical properties. An **atom** is the basic particle that makes up an element. Atoms of most elements can combine with other atoms. A **chemical bond** is the force that holds two atoms together. Atoms often combine to form **molecules**, which are larger particles made of two or more atoms held together by chemical bonds.

**When elements are chemically combined, they form compounds having properties that are different from those of the uncombined elements**. A **compound** is a pure substance made of two or more elements chemically combined in a set ratio. A compound may be represented by a **chemical formula**. A chemical formula shows the elements in the compound and the ratio of atoms. For example, the chemical formula for carbon dioxide is  $\text{CO}_2$ . In carbon dioxide, there are always two oxygen atoms to every one carbon atom.

Elements and compounds are pure substances, but most of the materials you see every day are not. Instead, they are mixtures. A **mixture** is made of two or more substances that are together in the same place, but are not chemically combined. Mixtures differ from compounds in two ways. **Each substance in a mixture keeps its individual properties. Also, the parts of a mixture are not combined in a set ratio.**

A mixture can be heterogeneous or homogeneous. In a **heterogeneous mixture**, you can see the different parts. The substances in a **homogeneous mixture** are so evenly mixed that you cannot see the different parts. A **solution** is an example of a homogeneous mixture. Air is a solution of nitrogen gas, oxygen gas, plus small amounts of other gases. Unlike compounds, mixtures are easily separated into their components. For example, iron filings can be easily removed from salt with a magnet.

# Concept Review Describing Matter Answers

**Daniel Chiras**



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*Matter* Prentice-Hall Staff, 1994      **Holt Chemistry** Holt Rinehart & Winston, 2003-01-24      **Concepts in Biology'** **2007 Ed. 2007 Edition** ,      *Ebook: Chemistry: The Molecular Nature of Matter and Change* Silberberg, 2015-01-16 Ebook  
*Chemistry The Molecular Nature of Matter and Change*      **Matter, Building Block of the Universe** , 1993      **Human Biology** Daniel Chiras, 2012 Written for the introductory human biology course the Seventh Edition of Chiras acclaimed text maintains the original organizational theme of homeostasis presented in previous editions to present the fundamental concepts of mammalian biology and human structure and function Chiras discusses the scientific process in a thought provoking way that asks students to become deeper more critical thinkers The focus on health and homeostasis allows students to learn key concepts while also assessing their own health needs An updated and enhanced ancillary package includes numerous student and instructor tools to help students get the most out of their course      **Chemistry: Core Concepts, 3rd Edition** Allan Blackman, Daniel Southam, Gwendolyn Lawrie, Natalie Williamson, Christopher Thompson, 2024-01-09 The third edition of *Chemistry Core Concepts* Blackman et al has been developed by a group of leading chemistry educators for students entering university with little or no background in chemistry Available as a full colour printed textbook with an interactive eBook code this title enables every student to master concepts and succeed in assessment Lecturers are supported with an extensive and easy to use teaching and learning package      *Matter and Energy* , 2005      *States of Matter, States of Mind* Allan F.M. Barton, 2021-08-29 *States of Matter States of Mind* is an easy to read introduction to the way the physical world is put together and stays together The book presents the fundamental ideas and particles of the makeup of the universe to enable understanding of matter and why it behaves in the way it does Written in an engaging manner the book explains some of the intricate details and grand schemes of life and the universe by making analogies with common everyday examples For example the recipe for a cake tells us nothing of how good the cake tastes but is a model of the food and a scientific model is no closer to the reality of the materials than a recipe is to the mouth watering flavor of the cake Illustrated with helpful cartoons this book provides a vast knowledge of atoms and atmospheres The first several chapters introduce terms and fundamental ideas while later chapters deal successively with particles and systems from the electron to the universe as a system Each new idea introduced builds upon the last A user friendly bibliography provides references for further reading      *Massage Therapy E-Book* Susan G. Salvo, 2019-02-28 Make the essential principles of massage therapy more approachable Covering massage fundamentals techniques and anatomy and physiology *Massage Therapy Principles and Practice 6th Edition* prepares you for success in class on exams and in practice settings The new edition of this student friendly text includes more than 700 images expanded information on the latest sanitation protocols critical thinking questions at the end of each chapter and updated pathologies which reflect what you will encounter in the field UPDATED pathologies ensure they are current and reflect what you will see in the field as a

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stuttgart is the capital of the bundesland of baden württemberg in germany with a population of approximately 632 000 in the immediate city 2017 and more than 5 2 million people in the metropolitan area 2013 stuttgart is the 6th largest city in germany stuttgart is known as a centre of mechanical and automobile engineering with the

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stuttgart covers an area of 207 35 km 2 80 sq mi and sits at elevation ranging from 207 m 679 ft above sea level by the neckar river to 549 m 1 801 ft on bernhartshöhe hill something rather unique in large german cities

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