## Acids, Bases & pH Review - Honors Chemistry

Directions: For all multiple choice questions, highlight the correct answer(s). Answer short answer questions in the box provided.

Part 1: Properties & definition of Acids & Bases

Complete the table below.

	ACIOS	BASES
Taxte	7	5
pH range	- 2	
Conducting electricity	70	
Neutralize		
Litmus	37	
Phenoiphthalein		

- 2. An indicator
  - a. Its a type of acid.
  - b. is a type of base.
  - c. has a neutral pH.
  - d. changes color depending on the pH.
- 3. Littinus and phenolofithalein are indicators. What does that meanly
  - Acids and bases "dissociate" in water to form H" (for acids) and OH (for bases).
     What does "dissociate" mean in this context?
  - Identify whether each substance is an acid or a base.
    - a. Hitter i.
    - B. HICHIONS
    - c. Alloomily:
    - d. NH + HyO + NH + OH; NH, is add):
    - H,SO, + H,O → H,O\* + HSO, : H,SO, is sónit
    - f. HNO, H\* + NO, : HNO, is admi-

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James E. Brady, Fred Senese

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