

Chemistry EOC Review Packet

Matter and Change: Pages 34-50

Procedures:

- Chemical change
- Chemical property
- Physical change
- Physical property
- Homogeneous mixture
- Heterogeneous mixture
- Compound
- Element
- Atom
- Isotope
- Ion
- Law of Conservation of Mass

Concept Questions:

- Write the definition of each of the following and explain what you mean:

Chemical change: A process in which a substance is transformed into one or more new substances with different chemical properties.

- Fill in the blank with the correct word(s) from the list below that best describes the change.

Change of Matter	Physical	Chemical
Boiling	Physical	Chemical
Freezing	Physical	Chemical
Burning	Physical	Chemical

- Which picture represents each state of matter?



Solid



Liquid



Gas

- What properties distinguish plasmas?

Plasma: A state of matter consisting of ionized gas, where electrons are separated from atoms.

- What properties distinguish supercritical?

Chemistry Matter And Change Review Packet Answers

**Petrogav International Oil & Gas
Training Center**

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the Petrogav International Oil & Gas Training Center text.

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