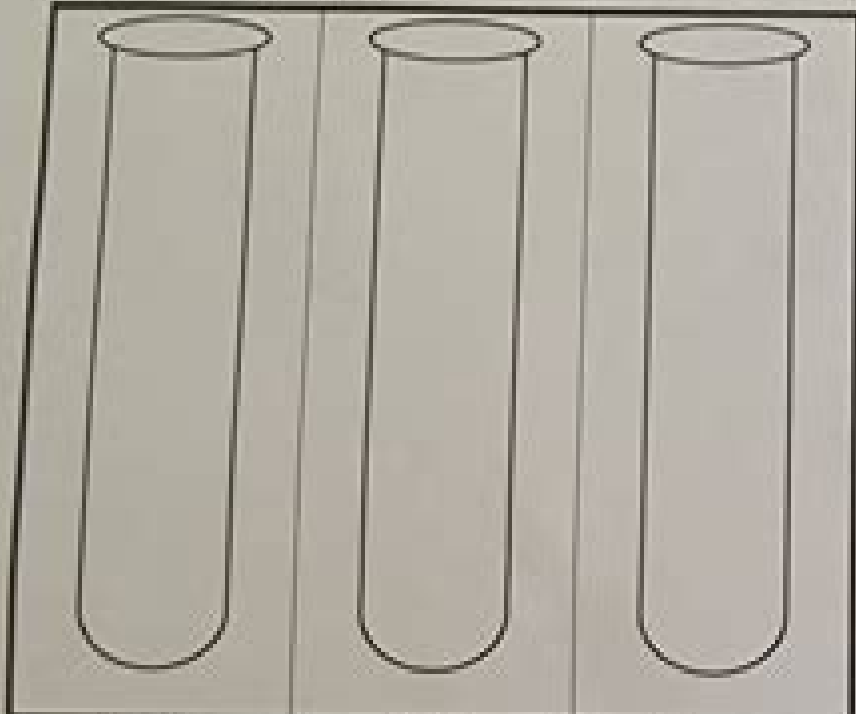


Chemistry 162 Nine Solutions Problem – Post Lab (Upload as an image to Canvas)

For each set of reactants shown: 1) Create small particle drawings of the separate reactant solutions and the solution after mixing. Maintain stoichiometry and include at least 2 H₂O molecules for (aq) mixtures 2) Write the balanced molecular reaction. 3) Write the balanced complete ionic equation (no cross-outs). 4) Write the balanced net ionic equation. Include phases for all species. See the lab manual for an example.

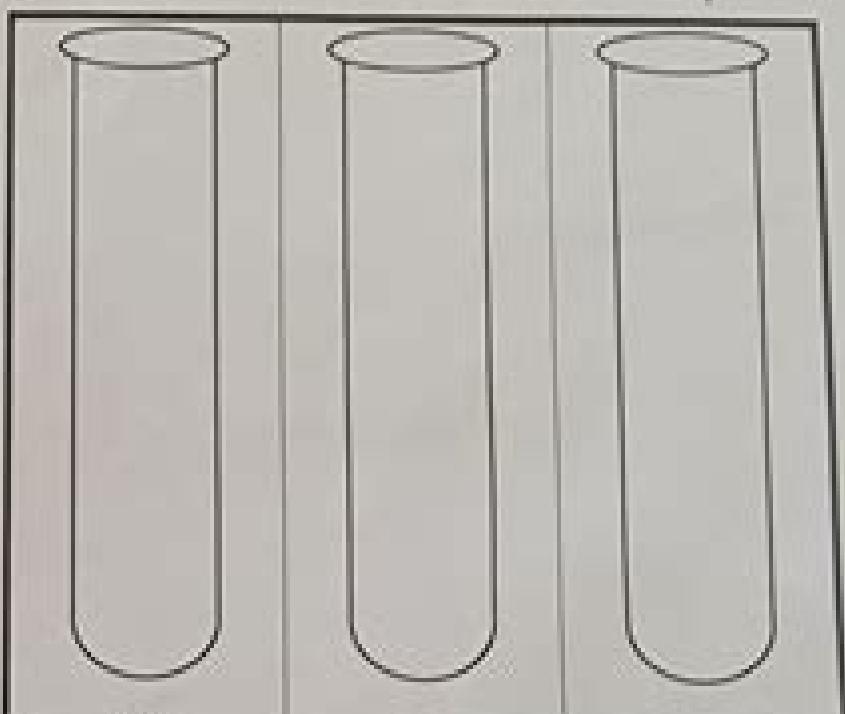


AgNO ₃	Na ₂ SO ₃	After mixing
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Balanced chemical equation:

Complete ionic equation

Net ionic equation



BaCl ₂	NaBr	After mixing
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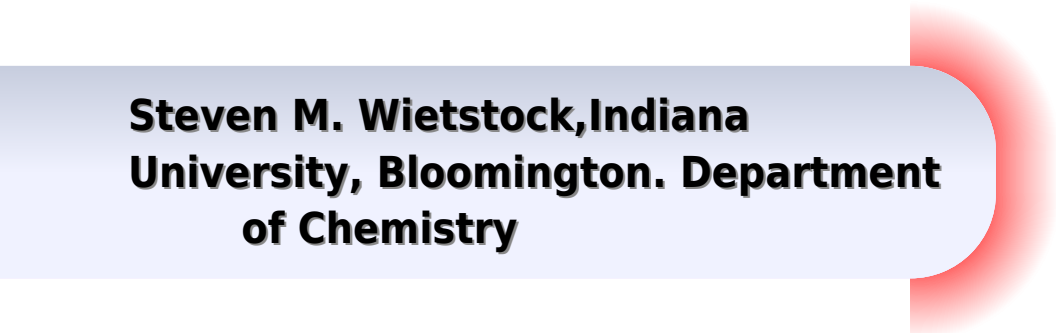
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**Steven M. Wietstock, Indiana
University, Bloomington. Department
of Chemistry**



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