

Exam 2

Total Points: 60

**PART 1: Multiple Choice questions: Q 1-24. (2 point each)**

- Chemical reactions (equations) show the following
  - Reactants
  - Products
  - Physical state of each (gas or liquid or solid)
  - ☒ All of the above
- The following is true for a mole
  - It is also known as Avogadro's number
  - It contains  $6.022 \times 10^{23}$  molecules
  - It contains mass equivalent to molecular weight
  - ☒ All of the above are true
- In balancing chemical equations, the coefficients are used to
  - Show number of moles
  - ☒ Show volume of each in liters
  - Show amount of each in grams
  - Show temperature at which the substance is stable
- $2\text{H}_2 + \text{O}_2 \longrightarrow 2\text{H}_2\text{O}$  means that
  - 2 grams of  $\text{H}_2$  g and 1 gram of  $\text{O}_2$  gas combine to form 2 grams of water.
  - ☒ 2 moles of  $\text{H}_2$  g and 1 mole of  $\text{O}_2$  gas combine to form 2 moles of water.
  - 2 grams of  $\text{H}_2$  g and 1 gram of  $\text{O}_2$  gas combine to form 2 moles of water.
  - 2 molecules of  $\text{H}_2$  g and 1 molecule of  $\text{O}_2$  gas combine to form 2 molecules of water.
- $2\text{H}_2 + \text{O}_2 \longrightarrow 2\text{H}_2\text{O}$ . In this equation, correct method to calculate molecular mass of water is
  - ☒  $2[2(\text{mass of hydrogen}) + \text{mass of oxygen}]$
  - $2(\text{mass of hydrogen gas}) + \text{mass of oxygen gas}$
  - $2(\text{mass of hydrogen}) + \text{mass of oxygen}$
- One mole of  $\text{O}_2$  gas contains \_\_\_\_\_ number of molecules.

# Chemistry 1 La B Manual Exam 2

**Katrin Zwirgmaier**



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appendix a many of a digital system s subsystems take the form of a switching circuit figure 1 1

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