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Chemistry 1411 Practice Exam 1, Chapters 1-3

Show your calculations clearly on show work problems. A periodic table is also included. Good luck!

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1	China there	mauroubsert of	stignificant.	dignames in th	he following	numbers (3 pts)	äc.

1) 6.022 X 10²³ atoms/mol 2) 0.049001 g

a) 3, 5

b) 4, 2

c) 23, 7

d) 4, 5

A rectangular block with a length, width, and height of 2.55 cm, 2.55 cm, and 1.44 cm has a density of 0.850 g/cm³. What is the mass of this object in grams? (3 pts)
 a) 5.46 g
 b) 7.96 g
 c) 8.65 g
 d) 10.4 g

(3 pts) Isotopes are atoms that:

a) have different atomic numbers

b) have the same mass number

c) have the same number of protons but different numbers of neutrons

d) have different numbers of electrons

4. A general summary based on observed and measurable behavior in many different systems

is called a: (3 pts)
a) hypothesis

b) natural law

c) model.

d) theory

What is the number of protons, neutrons, and electrons in the iron-56 atom? (3 pts)

a) 56 protons; 26 neutrons; 24 electrons b) 26 protons; 30 neutrons; 24 electrons

c) 24 protons; 56 neutrons; 26 electrons d) 26 protons, 30 neutrons; 26 electrons

Which of the following pairs of elements is a <u>transition element</u> and a <u>halogen</u>? (3 pts)

a) W and Si

b) Cr and Ar

c) Au and Br

d) Ba and U

Which of the following compounds is molecular? (3 pts)

a) BarCla

b) NaOH

c) SFa

d) KNOs

Which of the following elements is a noble gas? (3 pts)

a) Mg

b) Xe

e) Pb

d) Ber

Chemistry 1411 Lab Answers

Huangqi Zhang

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