

chemistry stoichiometry questions and answers

University of Binawan

1. Complete combustion of ammonia gas requires oxygen gas with equality reaction :



With Assuming that the oxygen level in the air is 20%, then the volume of air needed to burn 4 liters of ammonia gas is...

- A. 5 liters
- B. 7 liters
- C. 10 liters
- D. 35 liters

Answer: D

Discussion

Equality reaction equivalent :



Volume gas oxygen = $(7/4) \times 4$ liters = 7 liters

Volume oxygen = 20% volume air

7 liters = 0.2 volume of air

volume udara = $7/0.2 = 35$ litres

2. Every 1 liter of nitrogen gas is appropriate finished react with 2.5 liters of oxygen gas to form 1 liter of nitrogen oxide gas . If the volume is measured at the same temperature and pressure , then formula molecule the nitrogen oxide is

- A. N_2O
- B. NO
- C. N_2O_3 _ _ _
- D. N_2O_5 _

Answer: D

Discussion

Equality reaction :



The number of N atoms before reaction = Number of N atoms after reaction

$$4 = 2x$$

$$x = 2$$

Number of O atoms before reaction = Number of O atoms after reaction

$$10 = 2y$$

$$y = 5$$

So, formula molecule The nitrogen oxide is N_2O_5

3. Propane gas (C_3H_8) is burned perfect with reaction :



For burn perfect every liters of propane gas, oxygen gas is required as much

- A. 1.0 liters
- B. 3.0 liters

Chemistry Stoichiometry Guide Answers

Disha Experts

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'Disha Experts' text.

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The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter

Resources in Education, 1995

A Visual Analogy Guide to Chemistry, 2e Paul A Krieger, 2018-02-01 A Visual Analogy Guide to Chemistry is the latest in the innovative and widely used series of books by Paul Krieger This study guide delivers a big picture view of difficult concepts and effective study tools to help students learn and understand the details of general organic and biochemistry topics A Visual Analogy Guide to Chemistry is a worthwhile investment for any introductory chemistry student

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