

UNIT 1: MATTER & MEASUREMENT

Experimental Design

Independent Variable = Manipulated condition
Dependent Variable = Responding variable
Controlled Variable = Must be same for all groups

As part of the trial, substances are mixed in 4 beakers at the same temperature, volume and pH, resulting in the following concentrations:



Independent Variable = HCl Concentration (molarity)
Dependent Variable = Reaction time
Controlled Variable = Temperature, Mg sample size

Physical vs. Chemical

Physical:

- * Substance retains its properties
- * Chemical formulas do not change
- * Usually Phase Change

Examples:



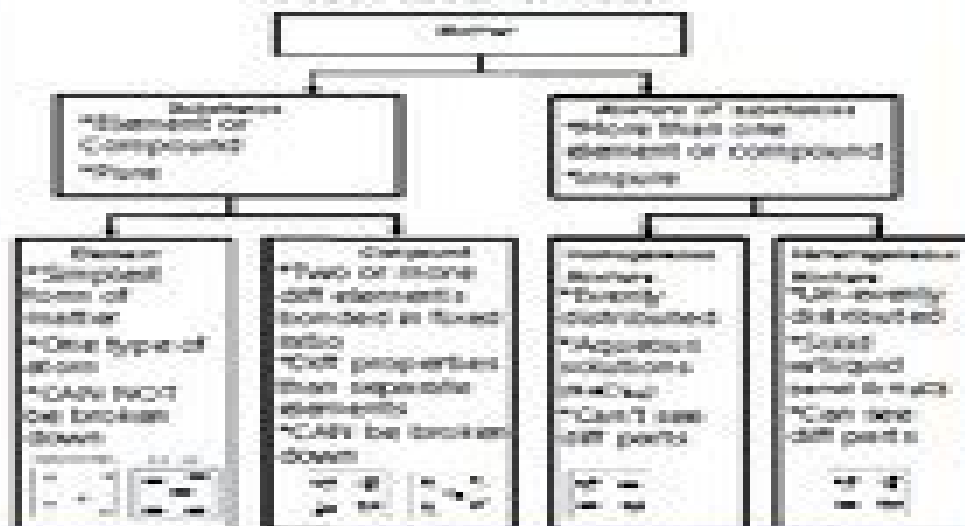
Chemical:

- * Properties do not change
- * Chemical formulas change
- * rust, burn, oxidize, combust, corrode

Examples:



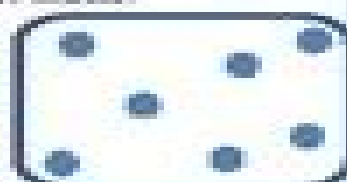
Classification of Matter



Phases of Matter

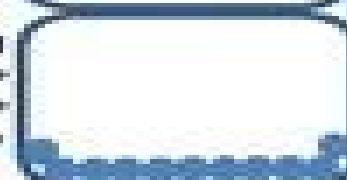
Gas

- NO definite volume
- NO definite shape
- Completely fills container
- Weak forces of attraction



Liquid

- definite volume
- NO definite shape
- flows and takes



Solid

- definite shape + volume
- regular geometric pattern
- vibrates in a fixed location



Physical Separation of Mixtures

Filtration

solid separated from liquid/solvents



key property: solubility (solid vs liquid/solvent)

Distillation

liquids with different PB heated to boil

key property: Boiling Point

Extraction

key property: Intermolecular Forces

Chromatography

different color ink dye separated by solubility

key property:

Molecular Polarity

Significant Figures

Atlantic Pacific Rule

Decimal point is present or significant



Multiply/Divide

Answer must have least sig figs in data

$$\begin{array}{r} 24.68 \\ \times 3.250 \\ \hline 79.226 \\ \hline 79.2 \end{array}$$

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C Cleary



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