

Periodic Trends

1. Explain the following trends:

- Atomic Radius
- Electronegativity
- Ionization Energy

2. Order the following elements by increasing atomic radius:

Li, Na, K, Rb, Cs, Fr

3. Order the following elements by increasing electronegativity:

F, Cl, Br, I, At

4. Order the following elements by increasing ionization energy:

Li, Na, K, Rb, Cs, Fr

5. Order the following elements by increasing metallic character:

Li, Na, K, Rb, Cs, Fr

6. Order the following elements by increasing nonmetallic character:

F, Cl, Br, I, At

7. Order the following elements by increasing atomic mass:

Li, Na, K, Rb, Cs, Fr

8. Order the following elements by increasing atomic number:

Li, Na, K, Rb, Cs, Fr

9. Order the following elements by increasing atomic weight:

Li, Na, K, Rb, Cs, Fr

10. Order the following elements by increasing atomic volume:

Li, Na, K, Rb, Cs, Fr

11. Order the following elements by increasing atomic density:

Li, Na, K, Rb, Cs, Fr

12. Order the following elements by increasing atomic melting point:

Li, Na, K, Rb, Cs, Fr

13. Order the following elements by increasing atomic boiling point:

Li, Na, K, Rb, Cs, Fr

14. Order the following elements by increasing atomic heat of fusion:

Li, Na, K, Rb, Cs, Fr

15. Order the following elements by increasing atomic heat of vaporization:

Li, Na, K, Rb, Cs, Fr

16. Order the following elements by increasing atomic heat of atomization:

Li, Na, K, Rb, Cs, Fr

17. Order the following elements by increasing atomic heat of hydration:

Li, Na, K, Rb, Cs, Fr

18. Order the following elements by increasing atomic heat of solution:

Li, Na, K, Rb, Cs, Fr

19. Order the following elements by increasing atomic heat of formation:

Li, Na, K, Rb, Cs, Fr

20. Order the following elements by increasing atomic heat of combustion:

Li, Na, K, Rb, Cs, Fr

Periodic Trends Test Study Guide

Section 1: Atomic Structure and Properties of the Periodic Table

1. What are the three main categories of elements in the periodic table?

- Metals
- Nonmetals
- Metalloids

2. List the following elements in order of increasing atomic radius: Li, Na, K, Rb, Cs, Fr.

3. List the following elements in order of increasing electronegativity: F, Cl, Br, I, At.

4. List the following elements in order of increasing ionization energy: Li, Na, K, Rb, Cs, Fr.

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20. List the following elements in order of increasing atomic heat of combustion: Li, Na, K, Rb, Cs, Fr.

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Pamela Mills,Amina El-Ashmawy



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