

Chemistry Final Exam Review

Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- ____ 1. The study of chemicals that, in general, do not contain carbon is traditionally called what type of chemistry?
a. bio c. physical
b. inorganic d. analytical
- ____ 2. Which of the following is NOT an example of matter?
a. air c. smoke
b. heat d. water vapor
- ____ 3. A golf ball has more mass than a tennis ball because it _____.
a. takes up more space c. contains different kinds of matter
b. contains more matter d. has a definite composition
- ____ 4. An example of an extensive property of matter is _____.
a. temperature c. mass
b. pressure d. hardness
- ____ 5. All of the following are physical properties of matter EXCEPT _____.
a. mass c. melting point
b. color d. ability to rust
- ____ 6. Which state of matter has a definite volume and takes the shape of its container?
a. solid c. gas
b. liquid d. both b and c
- ____ 7. Which of the following CANNOT be classified as a substance?
a. table salt c. nitrogen
b. air d. gold
- ____ 8. Which of the following is a heterogeneous mixture?
a. air c. steel
b. salt water d. soil
- ____ 9. Which of the following is a heterogeneous mixture?
a. vinegar in water c. oil and vinegar
b. milk d. air
- ____ 10. Separating a solid from a liquid by evaporating the liquid is called _____.
a. filtration c. solution
b. condensation d. distillation
- ____ 11. A substance that can be separated into two or more substances only by a chemical change is a(n) _____.
a. solution c. mixture
b. element d. compound
- ____ 12. The chemical symbol for iron is _____.
a. fe c. Fe
b. FE d. Ir

Chemistry Final Exam Review

Elsevier Inc, Jackie Koesterman



Chemistry Final Exam Review:

Final Exam Review: Arithmetic A. A. Frempong, Arithmetic covers Basic Definitions Terminology and Types of Numbers Writing Whole Numbers Using Numerals and Words Basic Operations and Properties Order of Operations and Evaluation of Arithmetic Expressions Rounding off Whole Numbers and Decimals Estimation Prime Numbers Divisibility Rules Prime Factorization Least Common Multiple LCM Operations on Fractions and Mixed Numbers Addition and Subtraction of Fractions Comparison of Fractions and Subtraction of Mixed Numbers Multiplication and Division of Fractions and Mixed Numbers Operations on Decimals Comparison of Decimals Complex Decimals Dividing Decimals Converting Fractions to Decimals Ratio and Proportion Proportion Problems Percent % and Calculations Involving Percent Averages Profit and Loss Areas and Perimeters Bar Line and Circle Pie Graphs Scientific Notation Measurements

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Chemistry II For Dummies John T. Moore, 2012-06-08 The tools you need to ace your Chemistry II course College success for virtually all science computing engineering and premedical majors depends in part on passing chemistry The skills learned in chemistry courses are applicable to a number of fields and chemistry courses are essential to students who are studying to become nurses doctors pharmacists clinical technicians engineers and many more among the fastest growing professions But if you're like a lot of students who are confused by chemistry it can seem like a daunting task to tackle the subject That's where Chemistry II For Dummies can help Here you'll get plain English easy to understand explanations of everything you'll encounter in your Chemistry II class Whether chemistry is your chosen area of study a degree requirement or an elective you'll get the skills and confidence to score high and enhance your understanding of this often intimidating subject So what are you waiting for Presents straightforward information on complex concepts Tracks to a typical Chemistry II course Serves as an excellent supplement to classroom learning Helps you understand difficult subject matter with confidence and ease Packed with approachable information and plenty of practice opportunities Chemistry II For Dummies is just what you need to make the grade

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OECD Reviews of Evaluation and Assessment in Education: Serbia Maghnouj Soumaya, Salinas Daniel, Kitchen Hannah, Guthrie Caitlyn, Bethell George, Fordham Elizabeth, 2020-03-20 This review developed in cooperation with UNICEF provides Serbia with recommendations to help strengthen its evaluation and assessment system to focus on support for student learning It will be of interest to Serbia as well as other countries looking to make more effective use of their evaluation and assessment system to improve quality and equity and result in better outcomes for all students

My Years in the Early Peace Corps: Ethiopia, 1965-1966 Sonja Krause Goodwin, 2021-09-17 The author describes her second year as a Peace Corps Volunteer teaching Chemistry in the Gondar Health College in Gondar Ethiopia a branch of Haile Selassie I University where she lectured taught laboratory courses and mixed solutions for her laboratory courses The students were not prepared for the classroom and she delves into her efforts to motivate them The college was also the local hospital and she describes her interactions with many physicians she met working at the hospital mostly expatriates She also describes her vacation travels during that time in and around Ethiopia and also to Kenya Tanzania and Uganda She visited several game parks and climbed Mt Kilimanjaro Goodwin also writes about her interactions fellow college teachers Peace Corps volunteers and Ethiopians She describes several instances of anti Peace Corps agitation in Ethiopia especially its effect on the local secondary school

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