

**Section 1.1** *continued*

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

19
18
Hydrogen
1.008
Helium

Atomic Mass	1.01
Atomic Number	18
Electron Configuration	18
Chemical Name	18
Chemical Symbol	18

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 19. It exists in the periodic table.
- _____ 20. It has no line periodic table.
- _____ 21. Group 1 elements.
- _____ 22. Elements that are strong oxidizing ability.
- _____ 23. Group 18 elements.

Column B

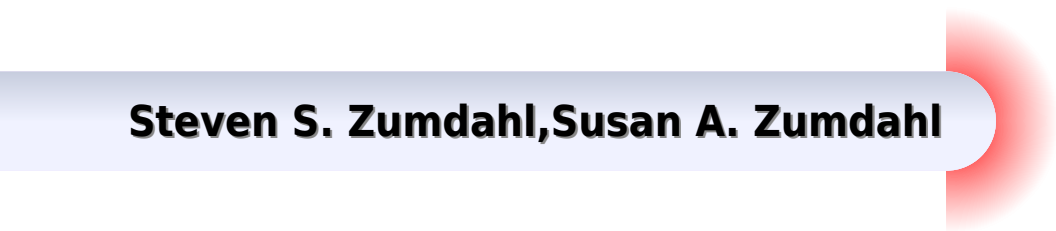
- a. noble
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write *yes* if the statement is true, if the statement is false, change the underlined word or phrase to make it true.

- _____ 24. There are two representative elements of elements.
- _____ 25. Most of the elements in the periodic table are metals.
- _____ 26. Group 1A elements except for hydrogen are metals in the solid state.
- _____ 27. Group 2A elements are the alkaline earth metals.
- _____ 28. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 29. Group 8A elements are inert nonmetallic elements known as noble gases.
- _____ 30. Alkaloids have properties of both metals and nonmetals.

Chemistry Chapter 6 Study Guide For Content Mastery Answers

Steven S. Zumdahl, Susan A. Zumdahl



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