

Periodic Trends

1. Explain the following trends:

- Atomic radius
- Electronegativity
- Ionization energy

2. Order the following elements by increasing atomic radius:

Li, Na, K, Rb, Cs, Fr

3. Order the following elements by increasing electronegativity:

F, Cl, Br, I, At

4. Order the following elements by increasing ionization energy:

Li, Na, K, Rb, Cs, Fr

5. Order the following elements by increasing atomic mass:

Li, Na, K, Rb, Cs, Fr

6. Order the following elements by increasing density:

Li, Na, K, Rb, Cs, Fr

7. Order the following elements by increasing melting point:

Li, Na, K, Rb, Cs, Fr

8. Order the following elements by increasing boiling point:

Li, Na, K, Rb, Cs, Fr

9. Order the following elements by increasing vapor pressure:

Li, Na, K, Rb, Cs, Fr

10. Order the following elements by increasing heat of fusion:

Li, Na, K, Rb, Cs, Fr

11. Order the following elements by increasing heat of vaporization:

Li, Na, K, Rb, Cs, Fr

12. Order the following elements by increasing heat of atomization:

Li, Na, K, Rb, Cs, Fr

13. Order the following elements by increasing heat of hydration:

Li, Na, K, Rb, Cs, Fr

14. Order the following elements by increasing heat of solution:

Li, Na, K, Rb, Cs, Fr

15. Order the following elements by increasing heat of formation:

Li, Na, K, Rb, Cs, Fr

16. Order the following elements by increasing heat of combustion:

Li, Na, K, Rb, Cs, Fr

17. Order the following elements by increasing heat of explosion:

Li, Na, K, Rb, Cs, Fr

18. Order the following elements by increasing heat of detonation:

Li, Na, K, Rb, Cs, Fr

19. Order the following elements by increasing heat of deflagration:

Li, Na, K, Rb, Cs, Fr

20. Order the following elements by increasing heat of ignition:

Li, Na, K, Rb, Cs, Fr

Periodic Trends Test Study Guide

Section 1: Atomic Structure and Properties of the Periodic Table

1. What are the three main categories of elements in the periodic table?

- Metals
- Non-metals
- Metalloids

2. List the following elements in order of increasing atomic radius: Li, Na, K, Rb, Cs, Fr.

3. List the following elements in order of increasing electronegativity: F, Cl, Br, I, At.

4. List the following elements in order of increasing ionization energy: Li, Na, K, Rb, Cs, Fr.

5. List the following elements in order of increasing atomic mass: Li, Na, K, Rb, Cs, Fr.

6. List the following elements in order of increasing density: Li, Na, K, Rb, Cs, Fr.

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19. List the following elements in order of increasing heat of deflagration: Li, Na, K, Rb, Cs, Fr.

20. List the following elements in order of increasing heat of ignition: Li, Na, K, Rb, Cs, Fr.

Chemistry Unit 10 Study Guide

Mr. Rohit Manglik



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