

Chemie

Hoofdstuk 1: Materie en stoffeigenschappen

1.1 Stoffeigenschappen

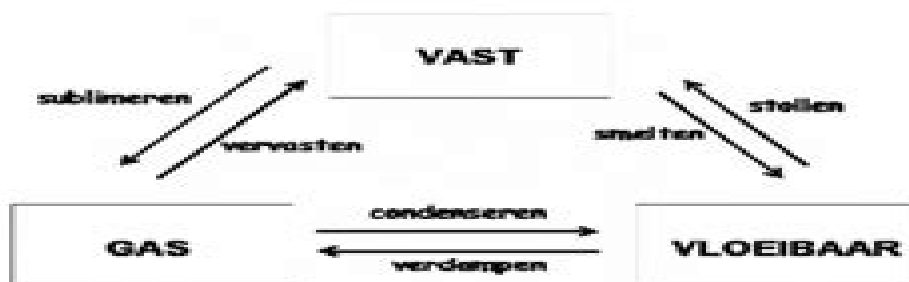
- **Materie**
= verzamelnaam voor stoffen, gekenmerkt door stoffeigenschappen
- **Stoffen**
 - Natuurlijke wijze
 - Chemische wijze

Stoffeigenschappen:

Direct waarneembaar	Meetbaar
• Uitzicht	• Hardheid
• Smaak en geur	• Massadichtheid
• Aggregatietoestand	• Kookpunt/ smeltpunt
	• Oplosbaarheid
	• Geleidbaarheid v warmte
	• Geleidbaarheid v elektrische stroom

1.2 Aggregatietoestand

- Deeltjesmodel
 - Beeld waarmee we verschijnselen proberen te verklaren
- Intermoleculaire kracht = cohesiekracht



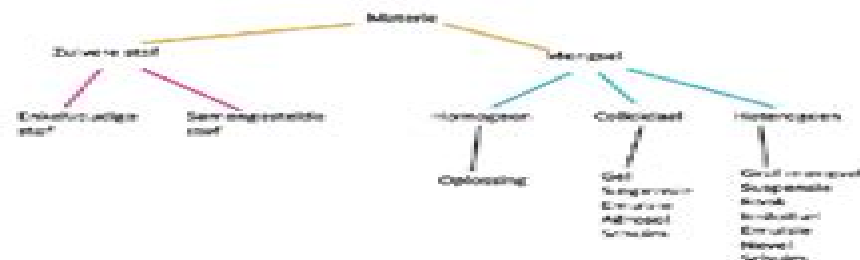
Aggregatietoestand / toestand	VASTE STOF	VLOEISTOF	GAS
Deeltjesmodel			
Ruimte tussen deeltjes	klein	groot	zeer groot
Aantrekkingskracht tussen deeltjes = cohesiekracht	sterk	minder sterk	zeer zwak

Rosterstructuur = deeltjes trillen

Deeltjes houden contact met elkaar = rollen over elkaar

Deeltjes bewegen vrij = botsen met elkaar

vast volume	bepaalde vorm	vast volume
geen vaste vorm en volume	sterke cohesiekracht	zwakke cohesiekracht
minder sterke cohesiekracht	neemt vorm aan van vat	samendrukbaar



Chemie Manipuleren Met Materie

F Rizvi



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