

molecular equation:



complete ionic equation:



net ionic equation:



spectator ions:



general reaction type:

Double replacement/displacement

#### VI. Heating Cu with Atmospheric $\text{O}_2$

appearance the unreacted Cu mesh:

shiny copper

appearance of the Cu mesh after heating and cooling:

dull dark brown

chemical equation:



general reaction type:

combination

Which species are oxidized?

Cu

Which species are reduced?

$\text{O}$

#### VII. Reacting $\text{CuSO}_4$ Solution with Steel Wool (Fe)

appearance of the steel wool (Fe) before reaction:

silver solid

appearance of the  $\text{CuSO}_4$  solution before reaction:

light blue

appearance of the steel wool (Fe) after the reaction has occurred:

dark brown solid

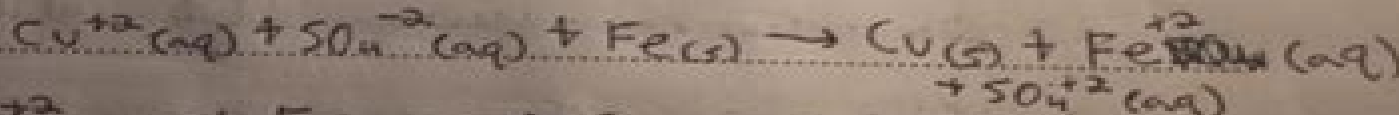
appearance of the  $\text{CuSO}_4$  solution after the reaction has occurred:

light blue liquid

molecular equation:



complete ionic equation:



net ionic equation:



spectator ions:



general reaction type:

single displacement

Which species are oxidized?

Fe

Which species are reduced?

$\text{Cu}^{+2}$

# Chemical Reactions Andequations Lab Manual Answers

**Olga I. Agapova**



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