Assessment Chemical Equations and Reactions Section Ouiz: Describing Chemical Reactions In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. 1. Which of the following is true regarding the law of conservation of a. It dictates that the number of molecules on each side of a chemical equation must be the same. b. It dictates that the number of atoms of each element must be the same on both sides of a chemical equation. c. It states that the mass of the reactants must remain constant in order for a chemical reaction to proceed. It does not apply to chemical reactions. 2. Which of the following may indicate that a chemical reaction has occurred? a. release of energy as light a color change c. gas bubble formation d. All of the above 3. Which of the following is a product in the reaction described by the word equation below? iron + copper(II) sulfate → iron(II) sulfate + copper iron(II) sulfate. b. comper c. iron. d. Both (a) and (b) 4. What does CH4 represent in the balanced chemical equation shown below?  $C + 2H_2 \rightarrow CH_4$ a. one atom of CH. b. five atoms of CH<sub>a</sub>. c.)one molecule of CH4 d. 1 g of CHa 5. Which of the following is not true regarding a properly written chemical equation? It must be balanced. b. It must contain proper chemical formulas. It must show the formation of a precipitate or a gas.

It should represent known data.

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