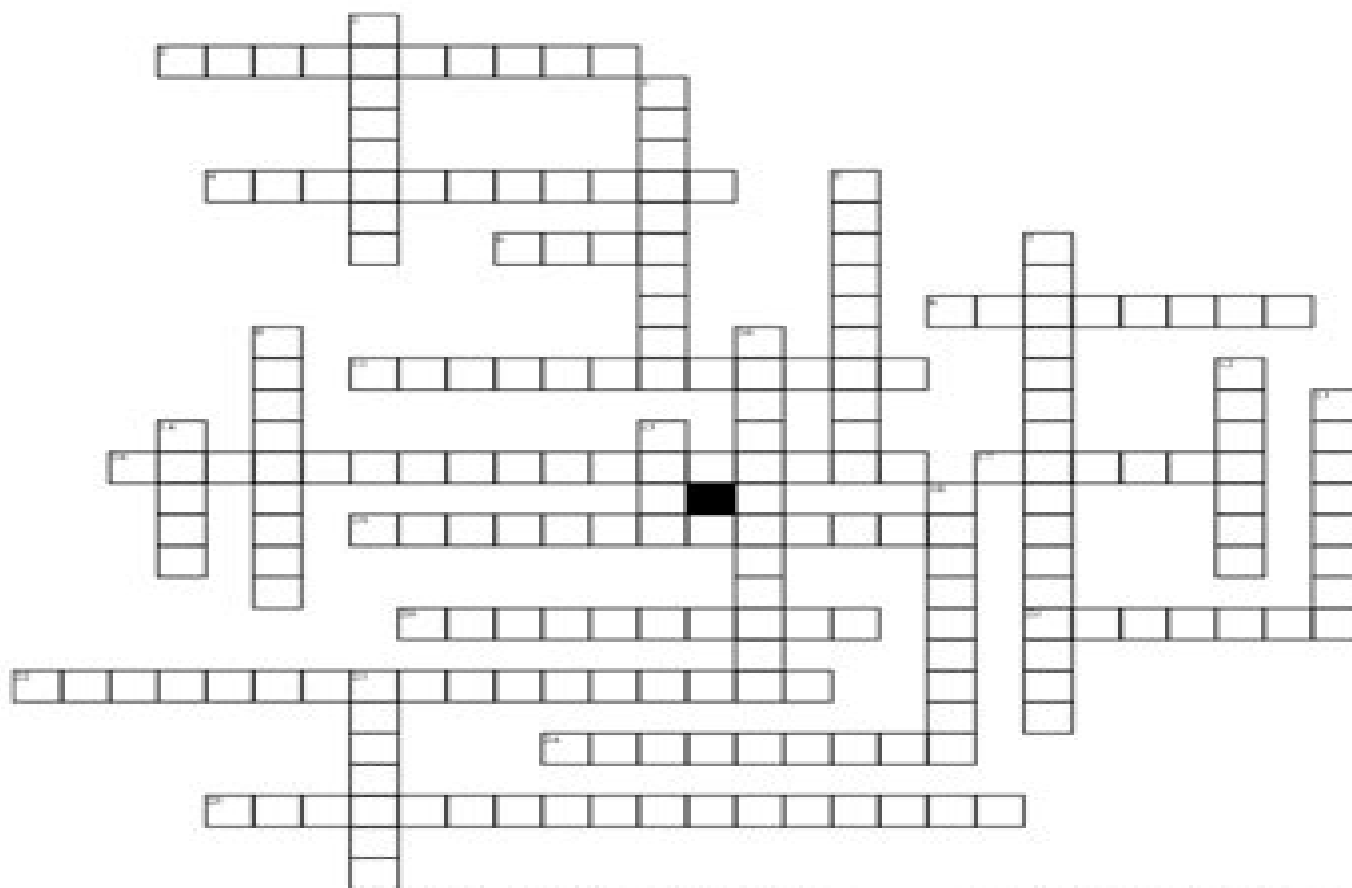


Chemical Equations



Across

2. an element will replace an element in a compound only if the single element is
 4. solid that forms as a result of a double replacement reaction
 6. compound with hydroxide ion is classified as a
 8. substance that speeds up a chemical reaction without being permanently changed
 11. numbers used to balance a chemical equation
 16. cations switch places to form two new compounds
 17. list of elements in order of reactivity is used with _____ replacement reactions
 19. in combustion reactions this gas is produced when oxygen is abundant

20. chart used to determine which product is a precipitate in double replacement reactions

21. proteins that acts as a biological catalyst
 22. one element replaces another element in a compound
 24. two or more substances combine to form a new compound
 25. Mass is not created nor destroyed during a chemical reaction

Down

1. chemical equation that has equal number of atoms of each element as reactants and products
 3. chemical reactions that release energy
 5. reaction of a hydrocarbon with oxygen to form water and carbon dioxide

7. minimum amount of energy required to start a chemical reaction

9. substance that slows down or stops a chemical reaction
 10. reaction in which one substance breaks down into simpler substances
 12. substance in a reaction represented by (aq)
 13. substance(s) formed as a result of a chemical reaction
 14. these numbers are required in compounds with transition metals
 15. any compound with hydrogen as its cation
 18. substance on the left of the reaction arrow
 23. measure of the disorder of a system

Chemical Formulas And Equations Crossword Puzzle

**David Martindill, Joanna
Haywood, Sheila Tarpey**



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all of them seemed to be vague at best and the standard of logical rigor seemed immensely inferior to what I could find in books on such other of the students in my first class subjects as calculus and fluid mechanics One who is now Prof F Gioia of the University of Naples once asked me a question which I have used here as Example 4 2 more than 20 years have gone by and I am still waiting for a more intelligent question from one of my students At the time that question compelled me to answer in a way I didn't like namely I'll think about it and I hope I'll have the answer by the next time we meet I didn't have it that soon though I did manage to have it before the end of the course

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This book has been written by a group of mathematicians and chemists whose common interest is in the complex dynamics of catalytic reactions Based on developments in mathematical chemistry a general theory is described that allows the investigation of the relationships between the kinetic characteristics of complex reactions and their detailed reaction mechanism Furthermore a comprehensive analysis is made of some typical mechanism of catalytic reactions in particular for the oxidation of carbon monoxide on platinum metals In fact the book presents three kinetics a detailed oriented to the elucidation of a detailed reaction mechanism according to its kinetic laws b applied with the aim of obtaining kinetic relationships for the further design of chemical reactors and c mathematical kinetics whose purpose is the analysis of mathematical models for heterogeneous catalytic reactions taking place under steady or unsteady state conditions

Index to Mathematical Problems, 1980-1984 Stanley Rabinowitz, 1992 A compendium of over 5 000 problems with subject keyword author and citation indexes

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