16. Given the chemical formula of an ionic compound, list at least three necessary steps to give the correct name of that compound. (If needed, use a chemical formula of a compound from the table above as an example in listing the naming steps.)

#### CHEMICAL BONDING POGIL

#### Activity 4 - Predicting the correct chemical fermula for leafs compounds formed from classic anions

#### Objective:

To learn how to credict the correct number of cations and anions in a simple salt.

#### The Models

Tables 3

Cathon	Anton	Chemical Formula	Compound Name
No.	CI	NaC3	sodiana chioride
58 m 1 2 2	CT	ZaC1,	mine elakorida
Nat	55.3	NagS	endium subfide
30.	N-2	K,N	potamicam nitride

#### Reviewing the Model

- 1. What is the charge on the zinc ion?
- 2. What is the charge on the nitride ion?
- 3. What is the charge on the chloride ion?
- 4. What is the charge on the ionic compound, sedium chicride?
- What is the charge on the ionic compound, sodium suffice?
- 6. How many potassium lons are present in K/N?
- 7. What does the "2" stand for in the formula for ZnCL?

#### Expliciting the Model

- B. Sodium chloride is NaCl, and sinc chloride ZnCl<sub>a</sub>. Why are there more chloride ions in the sinc compound?
- 9. Sodium chloride is NaCL and sodium suffide is Na.S. Why are there more sodium ions in the suffide compound?

# **Chemical Bonding Pogil Answer Key**

**Arshad Iqbal** 

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