

# Chemistry Review Questions

## Integrating Content, Inquiry and Reasoning

Review the following data from a calorimetry experiment to determine the heat of fusion of ice. After shaking off any excess water, several ice cubes were added to 99 g of warm water contained in a calorimeter. The initial temperature of the warm water was 46.8 °C. The ice-water mixture was stirred until the temperature reached a stable, minimum value, which was 1.1 °C. Any unmelted ice remaining at this point was immediately and carefully removed using tongs and the mass of the water in the calorimeter was measured—154 g.

known  
 most warm H<sub>2</sub>O → 99g  
 T<sub>i</sub> warm H<sub>2</sub>O → 46.8 °C  
 mass mixed → 154g - 99g = 55g  
 T<sub>fin</sub> (ice) → 0 °C

T<sub>f</sub> = 1.1 °C

1. Use the heat energy equation to calculate the amount of heat in joules released by the warm water as it cooled.

$$= mC_p(\Delta T) \\ = 99 \times 4.184 \times (46.8 - 1.1) \\ = 18929.7 \text{ J or } 18.93 \text{ kJ}$$

2. Based on the law of conservation of energy, what amount of heat was absorbed by the ice as it melted?

Heat absorbed → 18.93 kJ

3. Determine the amount of energy absorbed per gram of ice as it melted.

4. Calculate the heat of fusion (the heat required to melt ice) in units of kilojoules/mole.

5. The literature value for the heat of fusion of ice is 6.02 kJ/mole. What is the percent error for the experimentally determined heat of fusion?

6. When a mixture of ice and water originally at 0 °C is heated, the temperature remains constant (within experimental error) until all of the ice melts. Explain what happens to the heat energy that is absorbed during this time while the temperature does not change.

# Chemfax Ap Chemistry Laboratory 8 Answers

**Peter J. Mikulecky, Michelle Rose  
Gilman, Kate Brutlag**



## Chemfax Ap Chemistry Laboratory 8 Answers:

**Flinn Scientific Advanced Inquiry Labs for AP\* Chemistry** Flinn Scientific, 2013 AP Advantage Laboratory Investigations David W. Hostage, Martin Fossett, 2006-10-01 CliffsNotes AP Chemistry Bobrow Test Preparation Services, 2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter's end Two full length practice tests with detailed answer explanations are included in the book **Cracking the AP Chemistry Exam, 2015 Edition** Princeton Review, 2015-01-20 EVERYTHING YOU NEED TO SCORE A PERFECT 5 Equip yourself to ace the AP Chemistry Exam with The Princeton Review's comprehensive study guide including 2 full length practice tests thorough content reviews and targeted strategies for every section of the exam This eBook edition has been specially formatted for on screen viewing with cross linked questions answers and explanations We don't have to tell you how tough AP Chem is or how important a stellar score on the AP exam can be to your chances of getting into a top college of your choice Written by Princeton Review experts who know their way around chem Cracking the AP Chemistry Exam will give you Techniques That Actually Work Tried and true strategies to avoid traps and beat the test Tips for pacing yourself and guessing logically Essential tactics to help you work smarter not harder Everything You Need to Know for a High Score Comprehensive content review for all test topics Up to date information on the 2015 AP Chemistry Exam Engaging activities to help you critically assess your progress Practice Your Way to Perfection 2 full length practice tests with detailed answer explanations Practice drills at the end of each content chapter Review of important laboratory procedures and equipment *CHEMISTRY LAB INVESTIGATIONS* SAURYA. SINGH, 2021 **Cliffsnotes AP Chemistry** Angela Woodward Spangenberg, 2016 CliffsNotes AP Chemistry reflects recent changes to the AP Chemistry exam and includes subject review chapters and two full length practice AP Chemistry exams **Barron's AP Chemistry** Neil D. Jespersen, Pamela Kerrigan, 2016-03-01 Extensive test preparation for the AP Chemistry exam includes Six practice AP exams three diagnostic tests and three full length practice exams that are aligned with the upcoming new AP Chemistry exam All questions answered and explained A comprehensive subject review covering the structure of matter chemical bonding states of matter physical chemistry chemical reactions and all other test topics Study tips and test taking strategies This review book can be purchased alone or with an optional CD ROM that presents two additional full length practice tests with automatic scoring and fully explained answers *AP Chemistry For Dummies* Peter J. Mikulecky, Michelle Rose Gilman, Kate Brutlag, 2008-12-10 Gearing up for the AP Chemistry exam AP Chemistry For Dummies is packed with all the resources and help you need to do your very best This AP Chemistry study guide gives you winning test taking tips multiple choice strategies and topic guidelines as well as great advice on optimizing your study time and hitting the top of your game on test day This user friendly guide helps you prepare without perspiration by developing a pre test plan organizing your study time and getting the most out of your AP course You'll get help

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