

Finishing Part A

Use the chart below to determine the pH of each solution. Each pH indicator changes color based on the pH of the solution. Look at Methyl Red as an example. Methyl red will be red at and below a pH of 4.8. Additionally, methyl red will be yellow at or above a pH of 6.0. In between a pH of 4.8 and 6.0, methyl red will be orange (red and yellow make orange). Each pH indicator has a limited range. Use the combination of the indicators to narrow down the pH range for each solution. For example, consider a solution with unknown pH. Start with the indicators that change color around a pH of 7.0, which will be bromothymol blue and phenol red. Our bromothymol blue indicator is blue, so the pH must be greater than 7.6. Likewise, the phenol red indicator is red, so the pH must be greater than 8.0. However, our phenolphthalein solution is colorless, indicating a pH lower than 8.2. Therefore, we have determined that the solution must have a pH between 8.0 and 8.2.

13						12.8 Red
12						Alizarin yellow R
11					10.8 Pink	
10					Phenol phthalein	10.1 Yellow
9			7.6 Blue	8.8 Pink		
8			Bromothymol blue	Phenol red	8.2 Colorless	
7		6.0 Yellow				
6			4.8 Yellow	6.6 Yellow		
5	4.4 Yellow	Methyl red				
4		4.8 Red				
3	Methyl orange					
2	11 Red					
1						
0						

1. Record the experimental pH range for each solution below

Distilled Water	7.0
Tap Water	7.2
0.1M NaOH	11
0.1M NH_4OH	10.1
0.1M HAc	4.8
0.1M HCl	2.0
0.1M H_2SO_4	1.0

2. Calculate the theoretical pH for each solution and record it below

Distilled Water	
Tap Water	
0.1M NaOH	
0.1M NH_4OH	
0.1M HAc	
0.1M HCl	
0.1M H_2SO_4	

Chem 1212 Lab Manual

TD Snyder



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1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Chem 1212 Lab Manual
- Chapter 2: Essential Elements of Chem 1212 Lab Manual
- Chapter 3: Chem 1212 Lab Manual in Everyday Life
- Chapter 4: Chem 1212 Lab Manual in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Chem 1212 Lab Manual. The first chapter will explore what Chem 1212 Lab Manual is, why Chem 1212 Lab Manual is vital, and how to effectively learn about Chem 1212 Lab Manual.

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4. In chapter 3, this book will examine the practical applications of Chem 1212 Lab Manual in daily life. This chapter will showcase real-world examples of how Chem 1212 Lab Manual can be effectively utilized in everyday scenarios.

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6. In chapter 5, this book will draw a conclusion about Chem 1212 Lab Manual. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Chem 1212 Lab Manual.

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