

QUARTERLY EXAMINATION - 2023

11 - Std

CHEMISTRY

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Time : 3.00 hrs

Marks : 70

PART - A

I Answer all the questions.

15 X 1 = 15

- Which one of the following is used as a standard for atomic mass?
a) ${}_6\text{C}^{12}$ b) ${}_7\text{C}^{12}$ c) ${}_6\text{C}^{13}$ d) ${}_6\text{C}^{14}$
- Which one of the following represents 180g of water?
a) 5 moles of water b) 90 moles of water
c) $\frac{6.022 \times 10^{23}}{180}$ moles of water d) 6.022×10^{24} molecules of water
- Two electrons occupying the same orbital are distinguished by
a) Azimuthal quantum number b) Spin quantum number
c) magnetic quantum number d) orbital quantum number
- Which of the following element will have the highest electronegativity?
a) Chlorine b) Nitrogen c) Cesium d) Fluorine
- In a given shell the order of screening effect is
a) $s > p > d > f$ b) $s > p > f > d$ c) $f > d > p > s$ d) $f > p > s > d$

Chemistry Quarterly Exam First Answers

N Colangelo



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