

MULTIPLE CHOICE QUESTIONS

1. When potassium dichromate crystals are heated with hydrochloric acid, we get
(A) Oxygen (B) Hydrogen (C) Sulphur dioxide
(D) Hydrogen sulphide (E) Chlorine Ans : E
2. Which of the following elements has lowest melting point?
(A) Copper (B) Niobium (C) Molybdenum
(D) Tungsten (E) Mercury Ans : E
3. In acidic medium if potassium permanganate is used as an oxidising agent, we get
(A) MnO_4^{2-} (B) Mn^{2+} (C) MnO_2
(D) Mn_2O_3 (E) None of these Ans : B
4. Among the following outermost configurations of transition metals, which shows the highest oxidation state?
(A) $3d^3 4s^2$ (B) $3d^5 4s^1$ (C) $3d^6 4s^2$
(D) $3d^6 4s^2$ (E) $3d^7 4s^2$ Ans : C
5. Highest (+7) oxidation state is shown by
(A) Co (B) Cr (C) V
(D) Mn (E) Ni Ans : D
6. Potassium permanganate on heating gives
(A) $\text{K}_2\text{MnO}_4, \text{MnO}_2, \text{O}_2$ (B) $\text{MnO}_2, \text{O}_2, \text{KOH}$ (C) $\text{K}_2\text{MnO}_4, \text{O}_2$

Chemistry Multiple Choice Questions And Answers

Arshad Iqbal



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Table of Contents Chemistry Multiple Choice Questions And Answers

1. Understanding the eBook Chemistry Multiple Choice Questions And Answers
 - The Rise of Digital Reading Chemistry Multiple Choice Questions And Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Multiple Choice Questions And Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Multiple Choice Questions And Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Multiple Choice Questions And Answers
 - Personalized Recommendations
 - Chemistry Multiple Choice Questions And Answers User Reviews and Ratings
 - Chemistry Multiple Choice Questions And Answers and Bestseller Lists
5. Accessing Chemistry Multiple Choice Questions And Answers Free and Paid eBooks
 - Chemistry Multiple Choice Questions And Answers Public Domain eBooks
 - Chemistry Multiple Choice Questions And Answers eBook Subscription Services
 - Chemistry Multiple Choice Questions And Answers Budget-Friendly Options

6. Navigating Chemistry Multiple Choice Questions And Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Multiple Choice Questions And Answers Compatibility with Devices
 - Chemistry Multiple Choice Questions And Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Multiple Choice Questions And Answers
 - Highlighting and Note-Taking Chemistry Multiple Choice Questions And Answers
 - Interactive Elements Chemistry Multiple Choice Questions And Answers
8. Staying Engaged with Chemistry Multiple Choice Questions And Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Multiple Choice Questions And Answers
9. Balancing eBooks and Physical Books Chemistry Multiple Choice Questions And Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Multiple Choice Questions And Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Multiple Choice Questions And Answers
 - Setting Reading Goals Chemistry Multiple Choice Questions And Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Multiple Choice Questions And Answers
 - Fact-Checking eBook Content of Chemistry Multiple Choice Questions And Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
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