# CHAPTER 15 STUDY GUIDE

# **Energy and Chemical Change**

#### Section 15.1 Energy

In your textbook, read about the nature of energy.

In the space at the left, write true if the statement is true; if the statement is false, change the italicized word or phrase to make it true.

- Energy is the ability to do work or produce heat.

   The law of conservation of energy states that energy can be created and destroyed.
  - Chemical potential energy is energy stored in a substance because of its composition.
  - 4. Heat is a form of energy that flows from a warmer object to a cooler object.
  - A calorie is the amount of energy required to raise the temperature of one graw of pure water by one degree Celsius.
  - 6. A calorie is the SI unit of heat and energy.
  - The specific heat of a substance is the amount of heat required to raise the temperature of one gram of that substance by one degree Celsius.
  - 8. Kinetic energy is energy of motion.
  - Chemicals participating in a chemical reaction contain only potential energy.
  - 10. One nutritional Calorie is equal to 100 calories.
  - 11. One calorie equals 4.184 joules.
  - When a fuel is burned, some of its chemical potential energy is lost as heat.
  - To convert kilojoules to joules, divide the number of kilojoules by1000 joules/1 kilojoule.

#### Answer the following question. Show all your work.

 If the temperature of a 500.0-g sample of liquid water is raised 2.00°C, how much heat is absorbed by the water? The specific heat of liquid water is 4.184 J/(g-°C).

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