

### Kinetic Theory

- Gases are composed of \_\_\_\_\_, \_\_\_\_\_ particles called \_\_\_\_\_.
- Gas molecules are in \_\_\_\_\_.
- All \_\_\_\_\_ between particles are \_\_\_\_\_.
- The \_\_\_\_\_ of a gas display no \_\_\_\_\_ or \_\_\_\_\_ for one another.
- The \_\_\_\_\_ of the molecules is \_\_\_\_\_ to the \_\_\_\_\_ temperature of the gas.

### Ideal Gas

- Gas whose \_\_\_\_\_ conforms to the \_\_\_\_\_—it is \_\_\_\_\_.

### Gas Pressure:

Pressure = \_\_\_\_\_

Atmospheric Pressure - \_\_\_\_\_ the \_\_\_\_\_ the earth's \_\_\_\_\_ exerts due to its \_\_\_\_\_.

Barometer:

- Instrument used to measure \_\_\_\_\_.
- Invented by \_\_\_\_\_.

# Chemistry 901 Note Taking Guide

**D Kirk**



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