

**Chemistry 2
Semester 2 Final Exam
60 Points**

Name _____

Each question is worth 2 points. You need to indicate which TWO problems you would like to skip. Those problems will not be graded, and will not count for or against you. While SKIP simply refers to the problems as that I know it will not be graded. If you fail to follow directions, 10 points will be deducted automatically.

1. Name the following molecules:



Hydrogen sulfide acid

Copper (II) nitrate

Silicon dioxide

2. Give the chemical formula for the following compounds:

a. Copper (III) Phosphate

b. Phosphorous Acid

c. Fluorine trichloride

$CuPO_4$

H_3PO_3

Cl_3F

3. An unknown compound contains 17.9% Lithium, 26.8% Phosphorus, and 55.2% Oxygen. What is the empirical formula of the unknown compound?

$$17.9\% \text{ Li} / 1 \text{ mol} = 2.5 \text{ mol} \quad 3 \text{ Li}$$

$$26.8\% \text{ P} / 1 \text{ mol} = .2 \text{ mol} \quad 1 \text{ P}$$

$$55.2\% \text{ O} / 1 \text{ mol} = 3.4 \text{ mol} \quad 4 \text{ O}$$



4. A compound with the empirical formula C_2H_4 is found to have a molar mass of 112 g/mol. What is the molecular formula of the compound?

$$C_2H_4 = 28 \text{ g/mol}$$

$$\frac{112}{28} = 4 \rightarrow \boxed{C_{12}H_4}$$

5. Draw the following Lewis Structures:



Chemistry 1 Semester 2 Final Exam Review

**Michael Evans, Kay Lipson, Douglas
Wallace, Sue Avery**



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For over a decade Sandra McGuire has been acclaimed for her presentations and workshops on metacognition and student learning because the tools and strategies she shares have enabled faculty to facilitate dramatic improvements in student learning and success. This book encapsulates the model and ideas she has developed in the past fifteen years, ideas that are being adopted by an increasing number of faculty with considerable effect. The methods she proposes do not require restructuring courses or an inordinate amount of time to teach. They can often be accomplished in a single session, transforming students from memorizers and regurgitators to students who begin to think critically and take responsibility for their own learning. Sandra McGuire takes the reader sequentially through the ideas and strategies that students need to understand and implement. First she demonstrates how introducing students to metacognition and Bloom's Taxonomy reveals to them the importance of understanding how they learn and provides the lens through which they can view learning activities and measure their intellectual growth. Next she presents a specific study system that can quickly empower students to maximize their learning. Then she addresses the importance of dealing with emotion, attitudes, and motivation by suggesting ways to change students' mindsets about ability and by providing a range of strategies to boost motivation and learning. Finally, she offers guidance to faculty on partnering with campus learning centers. She pays particular attention to academically unprepared students, noting that the strategies she offers for this particular population are equally beneficial for all students. While stressing that there are many ways to teach effectively and that readers can be flexible in picking and choosing among the strategies she presents, Sandra McGuire offers the reader a step-by-step process for delivering the key messages of the book to students in as little as 50 minutes. Free online supplements provide three slide sets and a sample video lecture. This book is written primarily for faculty but will be equally useful for TAs, tutors, and learning center professionals. For readers with no background in education or cognitive psychology, the book avoids jargon and esoteric theory.

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