

- (iv) Fill in the blanks selecting the appropriate word from the given choice: [3]
- (a) In a covalent compound the bond is formed due to _____ of electrons (sharing / transfer)
- (b) A molecule which has a single lone pair of electrons _____. (NH_3 / H_2O)
- (c) Electrovalent compounds do not conduct electricity in their _____ state. (molten / solid)

Question 4

- (i) For each of the substances given below, what is the role played in the extraction of Aluminum. [2]
- (a) Cryolite
- (b) Graphite
- (ii) Calculate: [2]
- (a) A gas cylinder is filled with hydrogen and it holds 5 gms of gas. The same cylinder holds 85 gms of gas X under same temperature and pressure. Calculate the vapour density of gas X.
- (b) Give the empirical formula of CH_3COOH .
- (iii) The following questions are pertaining to the laboratory preparation of Hydrogen chloride gas. [3]
- (a) Write a balanced chemical equation for its preparation mentioning the condition required.
- (b) Why is concentrated Nitric Acid not used in the preparation of Hydrogen Chloride gas?
- (c) How is Hydrogen Chloride gas collected?
- (iv) Explain the following: [3]
- (a) Concentrated Nitric acid appears yellow when it is left standing in a glass bottle.
- (b) An inverted Funnel is used to dissolve Hydrogen Chloride gas in water.
- (c) All apparatus made of glass is used in the laboratory preparation of Nitric acid.

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