

Periodic Table of the Elements

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
1 H 1.008																	2 He 4.003
3 Li 6.941	4 Be 9.012											5 B 10.81	6 C 12.01	7 N 14.01	8 O 16.00	9 F 18.99	10 Ne 20.18
11 Na 22.99	12 Mg 24.31											13 Al 26.98	14 Si 28.09	15 P 30.97	16 S 32.06	17 Cl 35.45	18 Ar 39.95
19 K 39.10	20 Ca 40.08	21 Sc 44.96	22 Ti 47.88	23 V 50.94	24 Cr 52.00	25 Mn 54.94	26 Fe 55.85	27 Co 58.93	28 Ni 58.71	29 Cu 63.55	30 Zn 65.38	31 Ga 69.72	32 Ge 72.64	33 As 74.92	34 Se 78.96	35 Br 79.90	36 Kr 83.80
37 Rb 85.47	38 Sr 87.62	39 Y 88.91	40 Zr 91.22	41 Nb 92.91	42 Mo 95.94	43 Tc [98]	44 Ru 101.07	45 Rh 102.91	46 Pd 106.42	47 Ag 107.87	48 Cd 112.41	49 In 114.82	50 Sn 118.71	51 Sb 121.76	52 Te 127.60	53 I 126.91	54 Xe 131.29
55 Cs 132.91	56 Ba 137.33	57 La 138.91	58 Ce 140.12	59 Pr 140.91	60 Nd 144.24	61 Pm [145]	62 Sm 150.36	63 Eu 151.96	64 Gd 157.25	65 Tb 158.93	66 Dy 162.50	67 Ho 164.93	68 Er 167.26	69 Tm 168.93	70 Yb 173.05	71 Lu 174.97	72 Hf 178.49
77 Fr [223]	78 Ra [226]	79 Ac [227]	80 Rf [261]	81 Db [262]	82 Sg [266]	83 Bh [264]	84 Hs [277]	85 Mt [268]	86 Uun [289]	87 Uuu [288]	88 Uub [287]	89 Uuq [285]					

KEY

Atomic Mass → 12.011

Symbol → C

Atomic Number → 6

Electron Configuration → 2-4

Selected Oxidation States → -4, +2, +4

Relative atomic masses are based on $^{12}\text{C} = 12.000$

Nuclei Mass numbers in parentheses are mass numbers of the most stable or common isotopes.

Handwritten: Halogens

Handwritten: Alkali

Handwritten: Alkaline Earth

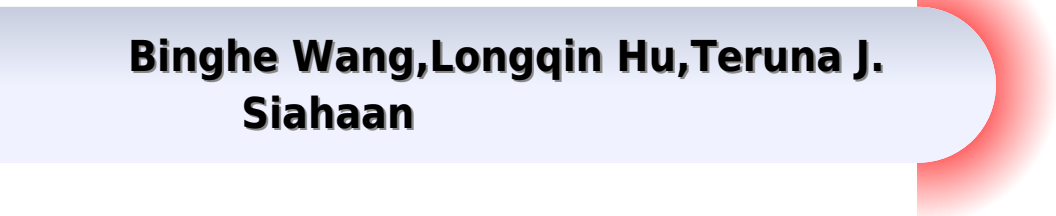
*Omit the presence of (2-8) for elements 72 and above

*The systematic names and symbols for elements of atomic numbers above 100 will be used until the approval of final names by IUPAC.

73 Ce 140.12	74 Pr 140.91	75 Nd 144.24	76 Pm [145]	77 Sm 150.36	78 Eu 151.96	79 Gd 157.25	80 Tb 158.93	81 Dy 162.50	82 Ho 164.93	83 Er 167.26	84 Tm 168.93	85 Yb 173.05	86 Lu 174.97
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**Binghe Wang, Longqin Hu, Teruna J.
Siahaan**



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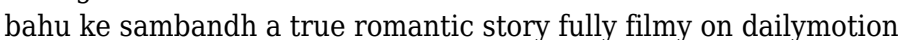
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