

Naming Compounds Practice Problems Use the periodic table located on the back of this page.

Name the following compounds:

- 1) FeBr_2 **iron(II) bromide**
- 2) Cu_2S **copper(I) sulfide**
- 3) Al_2O_3 **aluminum oxide**
- 4) Na_2S **sodium sulfide**
- 5) Fe_2O_3 **iron(III) oxide**
- 6) K_2SO_4 **potassium sulfate**
- 7) Al_2S_3 **aluminum sulfide**
- 8) BaF_2 **barium fluoride**
- 9) Ca_3N_2 **calcium nitride**
- 10) Ba_3P_2 **barium phosphide**
- 11) Ba_3N_2 **barium nitride**

Write the formulas for the following compounds:

- 1) sodium chloride **NaCl**
- 2) ammonium nitrate **NH_4NO_3**
- 3) lithium oxide **Li_2O**
- 4) ammonium phosphate **$(\text{NH}_4)_3\text{PO}_4$**
- 5) silver chloride **AgCl**
- 6) potassium nitrate **KNO_3**
- 7) magnesium chloride **MgCl_2**
- 8) calcium nitrate **$\text{Ca}(\text{NO}_3)_2$**
- 9) sodium hydroxide **NaOH**
- 10) magnesium nitrate **$\text{Mg}(\text{NO}_3)_2$**

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