#### Study Guide: Honors Chemistry Midterm 2025

Answer the following questions in separate sheets of paper. For answers requiring explanation, explain! For calculations, show <u>all</u> work! This study guide is graded!

#### \*\*\*Dates of the midterm\*\*\*

A Days: Wednesday, January 8 B Days: Friday, January 10

#### \*\*\*Dates of midterm corrections\*\*\*

A Days: Monday, January 13 B Days: Tuesday, January 14

#### **Physical and Chemical Properties**

- All matter has what two characteristics (that distinguish it from energy)?
   Have mass and takes up space (volume).
- 2) Intensive physical property versus extensive physical property
  - a. What is the difference between the two? Intensive properties do not depend on the amount of substance present and extensive properties depend on the quantity or the amount of substance.
  - Give two examples of each.
     Density and boiling point for Intensive.
     Mass and volume for Extensive.
  - c. Explain why only intensive physical properties can be used to identify an unknown object. Intensive properties are unique to a material and remain constant regardless of how much of the material is present. In contrast, extensive properties like mass and volume depend on the sample size and cannot uniquely identify a material.
- 3) What is the difference between a physical change and a chemical change? A physical change affects the substance's appearance, state, or properties, but its chemical composition remains unchanged and chemical change involves a reaction that alters the substance's chemical composition, resulting in the formation of new substances with different properties.
- 4) Label each of these as Physical (P) or Chemical (C) changes:

A. Food going bad (C)
B. Gold melting (P)
C. Salt dissolving (P)
D. Wood burning (C)
E. Paper tearing (P)
F. Water vaporizing (P)

- Density
  - a. What is density?

Density is a physical that measures how much mass is contained in each volume.

# **Chemistry Honors Midterm Study Guide Answers**

**Janet Lenore Harris** 

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