

# Chemistry 530 Midterm Review

## Exam on Monday Dec 16 at 8:30

### Section I = 75 pts

- Consists of 75 multiple choice questions worth 1 pt. each
- Questions cover all material covered during the second semester
- Questions will be similar to multiple choice problems in the book and review guides on-line
- Practicing the review guides is the best approach; answers will be posted on-line

### Section II = 95 pts

- Consists of 19 short answer problems totaling 95 points
- Questions cover the whole semester
- Examples of types of questions
  - Metric Conversions
  - Volume Calculations
  - Percent Composition
  - Atomic Mass
  - Electron configuration
  - Nomenclature
  - Balancing Equations
  - Predicting the products of a reactions
  - Identifying types of reactions
  - Mole Conversions
  - Molar Mass of a compound
  - Molar Mass of a gas at STP
  - Density of a gas at STP
  - Empirical & Molecular Formula

### Study approach

- First, read through your notes. Once thoroughly and once more quickly.
- Go through the review packet and make sure you are comfortable with all of the terms and how to do problems
- Do the review packets for help with multiple choice
- Do questions on old tests for practicing written problems
- Make sure you have practiced AT LEAST one of the types of example problems

### Chapter 1 – Introduction to Chemistry

- Chemistry: the branch of science that studies the composition and properties of matter.
- Scientific Method:
  - Observation
  - Hypothesis
  - Controlled Experiment
- Theory versus Natural Law

### Prerequisite Science Skills

- Rules for Significant Figures:
  - 1) All non-zero digits are significant.
  - 2) Zeros are significant only when they are *between* two non-zero digits or if they occur *after* a decimal point and *after* a non-zero digit.
  - 3) In *multiplication or division*, you cannot end up with more significant figures than you started with.
  - 4) In *addition or subtraction*, you cannot have more digits to the *right* of the decimal than you started with.
- Exponential numbers
- Scientific Notation =  $D.DD \times 10^n$  where  $1.00 \leq D.DD \leq 9.99$

# Chemistry Honors Midterm Study Guide Answer

**Marquis Who's Who**



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Darrell D. Ebbing, Larry K. Krannich, Joan I. Senyk, 1996

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