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1 kilo(unit) = 1000 (unit) 1 (unit) = 1000 milli

1 (unit) = 100 centi

micro = 10^6 nano = 10^9 $K = C + 273$ $C = K - 273$ H_2O FP = 273K

BP = 373K Density = mass/volume

Density of H_2O = 1 g/mLSig Figs: For addition & subtraction use least # of decimal places

/ For Multiplication & Division use least # of sig figs

Dalton's Atomic Theory: 1) all elements are composed of atoms

2) atoms of same element are identical

3) atoms can physically mix/chemically combine in whole #d ratios

4) chemical rxns happen; atoms don't change

Proton = pos., change = radioactivityNeutron = neutral, change = isotopeElectron = neg., change = ion

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