Stoichiometry Worksheet # 1

 Glucose is used as a source of energy by the human body. The overall reaction in the body is C6H12O6+6O2 → 6CO2 + 6H2O

Calculate the number of moles of oxygen needed to oxidize 12.5 g of C6H12O6.

Ammonia is synthesized from hydrogen and nitrogen according to the following equation.

$$N_2 + 3H_2 \rightarrow 2NH_3$$

If an excess of nitrogen is reacted with 3.41 moles of hydrogen gas, how many grams of ammonia (NH₃) can be produced?

3) Assume that in the decomposition of potassium chlorate, KCLO₃,

80.5 moles of O2 form. How many grams of potassium chlorite, the other product would be formed?

4) In a single displacement reaction,

9.23 moles of aluminum react with excess hydrochloric acid. How many moles of hydrogen will be produced?

5) The compound "Cisplatin" PtCl₂(NH₂)₂, has been found to be effective in treating some types of cancer. It can be synthesized using the following reaction

$$K_2PtCl_4 + 2NH_3 \longrightarrow 2KCl + PtCl_2(NH_3)_2$$

- A. How many moles of "cisplatin" can be produced from 2.50 g K2PtCl4?
- B. How many moles of NH3 would be needed if 3.8 moles of K2PtCl4 were present?

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Susan McMurry

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web fructose and lactose intolerance and malabsorption testing the relationship with symptoms in functional gastrointestinal disorders aliment pharmacol ther 2013 pdf download

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web some nuts are also fructose and lactose free examples include almonds and brazil nuts which lack both sugars almond butter is also free of fructose and lactose other nuts such as cashews filberts or hazelnuts and macadamia nuts also lack lactose but contain trace amounts of fructose

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pylori infection within these fourteen patients were lactose intolerant three showed fructose malabsorption and six