

ATOMIC STRUCTURES

MULTIPLE CHOICE QUESTIONS

1. The wave function ψ of 2s – orbital is given by

$$\psi_{2s} = \frac{1}{2\sqrt{2\pi}} \left(\frac{1}{a_0} \right)^{3/2} \left(2 - \frac{r}{a_0} \right) e^{-r/2a_0}$$

At $r = r_0$, radial node is formed. Which of the following is correct?

- (a) $r_0 = 2a_0$ (b) $2r_0 = a_0$
(c) $r_0 = 3a_0$ (d) None of these is correct

Explanation: (a)

At $r = r_0$, $\psi_{2s}^2 = 0$

It is possible only when $2 - \frac{r}{a_0} = 0$

$\therefore r_0 = 2a_0$

2. Which of the following set of quantum numbers belong to highest energy?

- (a) $n = 2, l = 1, m = 0, s = +\frac{1}{2}$
(b) $n = 3, l = 0, m = 0, s = +\frac{1}{2}$
(c) $n = 4, l = 0, m = 1, s = +\frac{1}{2}$
(d) $n = 4, l = 1, m = 1, s = +\frac{1}{2}$

Explanation: (d)

Chemistry Atomic Theory Questions

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Table of Contents **Chemistry Atomic Theory Questions**

1. Understanding the eBook **Chemistry Atomic Theory Questions**
 - The Rise of Digital Reading **Chemistry Atomic Theory Questions**
 - Advantages of eBooks Over Traditional Books
2. Identifying **Chemistry Atomic Theory Questions**
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a **Chemistry Atomic Theory Questions**
 - User-Friendly Interface
4. Exploring eBook Recommendations from **Chemistry Atomic Theory Questions**
 - Personalized Recommendations
 - **Chemistry Atomic Theory Questions** User Reviews and Ratings
 - **Chemistry Atomic Theory Questions** and Bestseller Lists

5. Accessing Chemistry Atomic Theory Questions Free and Paid eBooks
 - Chemistry Atomic Theory Questions Public Domain eBooks
 - Chemistry Atomic Theory Questions eBook Subscription Services
 - Chemistry Atomic Theory Questions Budget-Friendly Options
6. Navigating Chemistry Atomic Theory Questions eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Atomic Theory Questions Compatibility with Devices
 - Chemistry Atomic Theory Questions Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Atomic Theory Questions
 - Highlighting and Note-Taking Chemistry Atomic Theory Questions
 - Interactive Elements Chemistry Atomic Theory Questions
8. Staying Engaged with Chemistry Atomic Theory Questions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Atomic Theory Questions
9. Balancing eBooks and Physical Books Chemistry Atomic Theory Questions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Atomic Theory Questions
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Atomic Theory Questions
 - Setting Reading Goals Chemistry Atomic Theory Questions
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Atomic Theory Questions
 - Fact-Checking eBook Content of Chemistry Atomic Theory Questions
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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