

Answers to examination-style questions

Answers	Marks	Examiner's tips
1 a) i) 2	1	
ii) 1	1	
iii) 0	1	It is easier to calculate the order with respect to B first in this example.
b) $k = \frac{\text{rate}}{[D]^2[E]}$ or $\frac{8.36 \times 10^{-4}}{(0.84)^2(1.16)}$ $= 1.02 \times 10^{-3} \text{ mol}^{-2} \text{ dm}^6 \text{ s}^{-1}$	3	Make sure you can rearrange formulae. The units are worth a mark so do not leave them out. Answers from 1.02×10^{-3} to 1.05×10^{-3} are acceptable to take account of rounding up in the numbers during the calculation.
2 a) i) Expt 2 2.68×10^{-4} Expt 3 10.72×10^{-4} Expt 4 2.08×10^{-3}	3	Fewer than 2 dp here will be penalised. This gives 3 significant figs in answer. Accept 10.7×10^{-4}
ii) $k = \frac{\text{rate}}{[X]^2}$ or $\frac{2.68 \times 10^{-4}}{(1.20 \times 10^{-3})^2}$ $= 186 \text{ mol}^{-1} \text{ dm}^3 \text{ s}^{-1}$	3	Units are worth a mark so don't leave them out. In this case candidates often do part ii) to calculate k before part i).
b) increases exponentially, i.e. 	1	This is quite hard so a straight line like this was allowed:  but not a curve like this: 

Chemistry A2 Nelson Thornes Answers

Ted Lister, Janet Renshaw



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Make the Grade in AS and A2 Chemistry George Facer, 2003 *Make the Grade* offers comprehensive exam support for AS and A2 Chemistry Part of the Nelson Advanced Science series it provides activities and questions for use throughout the course with exam questions including synoptic questions to help students fully prepare for examinations **A-level Chemistry** E. N. Ramsden, 2000 Each topic is treated from the beginning without assuming prior knowledge Each chapter starts with an opening section covering an application These help students to understand the relevance of the topic they are motivational and they make the text more accessible to the majority of students Concept Maps have been added which together with Summaries throughout aid understanding of main ideas and connections between topics Margin points highlight key points making the text more accessible for learning and revision Checkpoints in each chapter test students understanding and support their private study **Chemistry for Advanced Level Course** Ted Lister, Janet Renshaw, 2000 This course study guide is designed to complement New Understanding Chemistry for Advanced Level but it can be used with any other core textbook for AS and A Level Chemistry as well It aims to help further develop chemistry skills such as laboratory techniques mathematical methods and data handling The course study guide also provides outline solutions to a selection of questions and gives advice on answering all types of examination questions and support for Key Skills **Essential A2 Chemistry for OCR** Janet Renshaw, Ted Lister, 2004 *Essential A2 Chemistry for OCR* provides clear progression with challenging material for in depth learning and understanding Written by the best selling authors of New Understanding Chemistry these texts have been written in simple easy to understand language and each double page spread is designed in a contemporary manner Fully networkable and editable Teacher Support CD ROMs are also available for this series containing worksheets marking schemes and practical help **Nelson Modular Science** Paul Collison, 2002 The Nelson Modular Science series is made up of three books divided into single double and triple award modules presented in an accessible format Book 1 covers the six single award and one coursework modules Book 2 contains six double award modules and Book 3 covers the six triple award modules Each module is covered in self contained units This teacher's file includes practical support sheets and addresses Sc1 investigations Works sheets are provided to integrate the use of ICT throughout science Additional GCSE style questions and modular tests should enhance learning and recall of information **Physics for Advanced Level** Jim Breithaupt, 2000 This course study guide is to be used with New Understanding Physics for Advanced Level or other physics core textbooks It aims to help further develop physics skills such as laboratory techniques mathematical methods and data handling The course study guide also provides outline solutions to a selection of questions and gives advice on answering all types of examination questions and support for Key Skills **Advanced Chemistry for You** Lawrie Ryan, 2000 *Advanced Chemistry for You* has been carefully designed to be interesting and motivating to the AS A2 student with features that make it highly supportive of individual learning Written by an experienced author team with the

same straightforward approach as the successful New for You GCSE series *Calculations for A-level Chemistry* E. N. Ramsden, 2001 Aiming to match the various specifications this book gives explanations worked examples and practice in chemistry calculations It includes a comprehensive mathematics foundation section Work on formulae and equations the mole volumetric analysis and other key areas are included It is useful as a course book as well as for exam practice

Inorganic Chemistry James E. House, 2012-12-31 Inorganic Chemistry Second Edition provides essential information for students of inorganic chemistry or for chemists pursuing self study The presentation of topics is made with an effort to be clear and concise so that the book is portable and user friendly The text emphasizes fundamental principles including molecular structure acid base chemistry coordination chemistry ligand field theory and solid state chemistry It is organized into five major themes structure condensed phases solution chemistry main group and coordination compounds with several chapters in each There is a logical progression from atomic structure to molecular structure to properties of substances based on molecular structures to behavior of solids etc The textbook contains a balance of topics in theoretical and descriptive chemistry For example the hard soft interaction principle is used to explain hydrogen bond strengths strengths of acids and bases stability of coordination compounds etc Discussion of elements begins with survey chapters focused on the main groups while later chapters cover the elements in greater detail Each chapter opens with narrative introductions and includes figures tables and end of chapter problem sets This new edition features new and improved illustrations including symmetry and 3D molecular orbital representations expanded coverage of spectroscopy instrumental techniques organometallic and bio inorganic chemistry and more in text worked out examples to encourage active learning and to prepare students for their exams This text is ideal for advanced undergraduate and graduate level students enrolled in the Inorganic Chemistry course This core course serves Chemistry and other science majors The book may also be suitable for biochemistry medicinal chemistry and other professionals who wish to learn more about this subject area Concise coverage maximizes student understanding and minimizes the inclusion of details students are unlikely to use Discussion of elements begins with survey chapters focused on the main groups while later chapters cover the elements in greater detail Each chapter opens with narrative introductions and includes figures tables and end of chapter problem sets *New Understanding Chemistry for Advanced Level Third Edition* Ted Lister, Janet Renshaw, 2000 Matches the specifications of the Awarding Bodies AQA NEAB AEB OCR and Edexcel This accessible text includes frequent hints questions and examination questions providing support and facilitating study at home It features photographs and comprehensive illustrations with 3D chemical structures

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