Exercises

Following the rule you established in the last key question, write correct chemical formula for each of the following compounds

Compound	Formula
(a) Lithium chloride	L." /ce" -> L.CQ
(b) Magnesium iodide	Mg = / I - Mg Iz
(c) Strontium selenide	51'2/Se2 >> SrSe
(d) Rubidium fluoride	Rb1/F7 - RbF
(e) Lithium oxide	L:+1/0-2 -> L:20
(f) Sodium sulfide	Na" 15-2 -> Na25
(g) Potassium chloride	K+1/ce-1 -> Kce
(h) Calcium phosphide	Ca+2/P-3 - Ca2P2
(i) Barium oxide	Bc+2/0-2 > Bc0
(j) Aluminum sulfide	RQ+3/5-2 -> AD253

 Use your answers to the Key Questions and the Exercise Questions to draw a conclusion about the ratio of ions in two compounds if the elements in the compounds are from the same groups (example: aluminum oxide and aluminum sulfide; lithium chloride and potassium chloride).

Chemistry A Bonding Packet Answers

Gerhard Quinkert

Chemistry A Bonding Packet Answers:

Engineering Chemistry-I: Concepts and Applications Jit Chakraborty & Asimesh Dutta Gupta and Ravikanth Kamlekar, Engineering Chemistry I Concepts and Applications is a textbook that offers an exclusive coverage of the topics and proper explanation of concepts as per the present day and future needs of the students The book provides the theoretical Chapters 1 7 as well as practical Chapter 8 aspects of the paper Chemistry I BSC102 as per the latest AICTE curriculum It will be useful to not only the first year engineering and technology students of all streams but also the professors for guiding their students

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to describe and understand people and their mental lives This book explores how this observation plays out in the contexts of
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What do explanations of how an artifact fulfills its function look like and do they differ from explanations of natural systems

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