

Balancing and Reactions Test Unit 2 SCH3U

Name: _____

 K T C A **Assessment of Learning (Product)**
15 20 5 4

Multiple Choice: Circle the most correct answer. (10K)

1. Propane gas, C_3H_8 , burns in excess oxygen gas. When the equation for this reaction is correctly balanced and all coefficients are reduced to their lowest whole-number terms, the coefficient for H_2O is...

- A) 1 B) 2 C) 3 D) 4 E) 5 f) 6

2. State the type of reaction: $2 Na + 2 H_2O \rightarrow 2 NaOH + H_2$

- A) synthesis B) decomposition C) single displacement
D) double displacement E) combustion F) no reaction

3. Which of the following types of chemical reactions produces water and an ionic compound?

- A) single displacement B) neutralization C) decomposition D) synthesis

4. State the type of reaction: $Ca(NO_3)_2 + 2 HCl \rightarrow CaCl_2 + 2 HNO_3$

- A) synthesis B) decomposition C) single displacement
D) double displacement E) combustion F) no reaction

5. If chlorine gas is produced by single displacement, the other halogen in the reaction must be

- A) bromine B) iodine C) astatine D) fluorine

6. In what kind of reaction do two or more substances combine to form a new compound?

- A) decomposition reaction C) double-displacement reaction
B) single displacement D) synthesis reaction

7. A group whose atoms (M) undergo the reaction:



- A) group 1 B) group 2 C) group 13 D) group 14 E) group 15

8. When heated, a metal carbonate decomposes into a metal oxide and

- A) carbon. B) carbon dioxide C) oxygen. D) hydrogen.

9. Oxides of active metals, such as CaO , react with water to produce

- A) metal carbonates. B) metal hydrides. C) acids. D) metal hydroxides.

10. Which of the following statements is true about single-displacement reactions?

- A) They are restricted to metals. C) Two reactants produce two products.
B) They involve a single product. D) Any metal replaces any other metal.

Chemical Equations And Reactions Test Answers

Adrian Dingle



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