



Worksheet  
Chemical Bonding

CLASS : XI

**Multiple choice questions**

1. The pyramidal geometry is associated with:  
a)  $\text{CH}_4$   
b)  $\text{NH}_3$   
c)  $\text{H}_2\text{O}$   
d)  $\text{CO}_2$
2. Which one of them is the weakest?  
a) Ionic bond  
b) Covalent Bond  
c) Metallic bond  
d) Van der Waal's force
3. Among  $\text{LiCl}$ ,  $\text{BeCl}_2$ ,  $\text{BCl}_3$  and  $\text{CCl}_4$ , the covalent character varies as:  
a)  $\text{LiCl} < \text{BeCl}_2 < \text{BCl}_3 < \text{CCl}_4$   
b)  $\text{LiCl} > \text{BeCl}_2 > \text{BCl}_3 > \text{CCl}_4$   
c)  $\text{LiCl} < \text{BeCl}_2 < \text{BCl}_3 < \text{CCl}_4$   
d)  $\text{LiCl} > \text{BeCl}_2 > \text{BCl}_3 > \text{CCl}_4$
4. Which of the following contains no pi bond?  
a)  $\text{C}_2\text{H}_4$   
b)  $\text{C}_3\text{H}_8$   
c)  $\text{C}_2\text{H}_2$   
d)  $\text{N}_2$
5. Bond order of 1.5 is shown by :  
a)  $\text{O}_2^+$   
b)  $\text{O}_2^-$   
c)  $\text{O}_2^{2-}$   
d)  $\text{O}_2^+$
6. Which of the following molecules have trigonal geometry?  
a)  $\text{H}_2\text{O}$   
b)  $\text{NH}_3$   
c)  $\text{PCl}_3$   
d)  $\text{BF}_3$
7. The hybridization state of carbon in diamond, graphite and acetylene is  
a)  $sp^3$ ,  $sp^2$ ,  $sp$   
b)  $sp$ ,  $sp^3$ ,  $sp^2$   
c)  $sp^2$ ,  $sp^3$ ,  $sp$   
d)  $sp$ ,  $sp^2$ ,  $sp^3$
8. The H-O-H bond angle in water is  
a)  $104.5^\circ$   
b)  $120^\circ$   
c)  $107^\circ$   
d)  $180^\circ$

# Chemical Bond Multiple Choice Questions

**Arihant Experts**



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