CHEMICAL REACTIONS

Chapter Test B

A. Matching

Match each term in Column B with the correct description in Column A. Write the terrer of the correct term on the line.

Column B Column A ____ 1. a reaction in which an element or compound a. decomposition reaction reacts with oxygen, often producing energy in the form of heat or light 2. a reaction in which two or more substances b. activity series react to form a single substance 3. an equation that indicates only those particles c. spectator ions. that actually take part in the reaction. a substance that speeds up a reaction without d. balanced equation being used up 5. Ions that are not directly involved in a reaction. e. double-replacement reaction. 6. a reaction in which a single compound is f. catalyst broken down into two or more products 7. an equation in which each side has the same g. combustion reaction number of atoms of each element B. a reaction in which atoms of an element replace b. combination reaction. the atoms of a second element in a compound 9. a list of metals in order of decreasing reactivity L. netionic equation. 10. a reaction that involves an exchange of positive single-replacement reaction. ions between two compounds B. Multiple Choice Choose the best answer and write its letter on the line.

- 11. In the chemical equation 2H₂O₂(as) $\stackrel{\text{declay}}{\longrightarrow}$ 2H₂O(t) + O₂(s), the MnO₂ is a: a. reactant. c. spectator ion. b. product. d. catalyst. 12. When the equation Mg(t) + HCl(eq) → MgCl₂(eq) + H₂(g) is balanced.
- what is the coefficient for HCIR
 - b. 2

c. 3

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Venkata Rao Basa

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