



Organic Chemistry

Test Study Guide:

Ch.7 Enolates

- Aldehyde and Ketones
- Enolate Chemistry
- General Principles
- Aldol Condensation
- And More

CONCEPT SUMMARY

General Principles of Carbonyl Chemistry:

1. **α -Carbon Identification:** The carbon adjacent to the carbonyl carbon is referred to as the α -carbon, while the hydrogens attached to this α -carbon are known as α -hydrogens.
2. **Acidity of α -Hydrogens:** α -Hydrogens exhibit relatively acidic properties and can be effectively removed by strong bases.
3. **C-H Bond Weakening:** The presence of the electron-withdrawing oxygen in the carbonyl group contributes to the weakening of C-H bonds on α -carbons.
4. **Enolate Stability via Resonance:** Upon deprotonation, the resulting enolate can be stabilized through resonance with the carbonyl group.
5. **Reactivity of Ketones:** Ketones show reduced reactivity towards nucleophiles due to steric hindrance and destabilization of the α -carbanion.
6. **Impact of Alkyl Groups:** An additional alkyl group introduces crowding in the transition state, leading to increased energy and decreased stability of the carbanion due to electron density donation.

Points of Enolate Chemistry

1. Keto and Enol Forms

Aldehydes and ketones predominantly exist in the keto form ($C=O$), but can also exist in the enol form, characterized by a double bond and a hydroxyl group ($C=C-OH$).

2. Tautomerism

Tautomers are isomers that can be interconverted through the migration of a hydrogen atom and a double bond. The keto and enol forms are examples of such tautomers.

3. Formation of Enolates

The enol form can undergo deprotonation to yield an enolate, which serves as a strong nucleophile in chemical reactions.

4. Michael Addition Reaction

In the Michael addition mechanism, an enolate attacks an α,β -unsaturated carbonyl compound, resulting in the formation of a new carbon-carbon bond.

5. Kinetic vs. Thermodynamic Enolates

Kinetic enolates are favored under conditions of fast, irreversible reactions at lower temperatures using strong, sterically hindered bases.

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