

For each problem below, list each variable given and which one is unknown, choose the correct equation and plug in the numbers to solve for the unknown.

2. 234 kPa of Argon is at a temp of 255 K. If the temp changes to 100 K, what is the new pressure?

$$P_1 = 234 \text{ kPa}$$

$$T_1 = 255 \text{ K}$$

$$T_2 = 100 \text{ K}$$

$$P_2 = ?$$

$$100 \text{ K} \left(\frac{234 \text{ kPa}}{255 \text{ K}} \right) = \frac{P_2}{100 \text{ K}} (100 \text{ K})$$

$$P_2 = 91.765 \text{ kPa}$$

3. A 6.5 liter tank of methane gas has compressed gas in it so the pressure is 2500 kPa. What will the pressure of the gas be if the gas is put into a 3.25 liter tank?

$$V_1 = 6.5 \text{ L}$$

$$P_1 = 2500 \text{ kPa}$$

$$V_2 = 3.25 \text{ L}$$

$$P_2 = ?$$

$$\frac{(2500)(6.5)}{3.25} = \frac{P_2 (3.25)}{3.25}$$

$$P_2 = 5000 \text{ kPa}$$

4. Neon gas at a pressure of 25 kPa occupies 444 liters at 333 K. What is the pressure of the gas at 200 K in a volume of 30 liters?

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Ideal Gas Law 51 Success Secrets - 51 Most Asked Questions on Ideal Gas Law - What You Need to Know Philip Hicks, 2014-10-07 It's a brand new Ideal gas law world There has never been a Ideal gas law Guide like this It contains 51 answers much more than you can imagine comprehensive answers and extensive details and references with insights that have never before been offered in print Get the information you need fast This all embracing guide offers a thorough view of key knowledge and detailed insight This Guide introduces what you want to know about Ideal gas law A quick look inside of some of the subjects covered Atmospheric thermodynamics Overview Thermodynamic instruments Thermodynamic meters Glossary of engineering I Idealization Limits on use Perfect gas Stoichiometry Water vapor Water vapor and dry air density calculations at 0 C Equipartition theorem Perfection Physics and chemistry Glossary of chemistry terms U Fusion energy 1960s Timeline of low temperature technology 19th century Gas Avogadro's law Hot air balloon List of multiple discoveries 17th century Amount of substance Equation of state Overview Explosive Volume of products of explosion Aerodynamics Conservation laws Van der Waals equation Validity Equipartition of energy Gas Physical characteristics Gas meter Flow measurement calculations Mass flow sensor Chamber pressure Importance in Firearm Maintenance Weather forecasting How models create forecasts Timeline of hydrogen technologies 1800s Pressure Pressure of an ideal gas Compressible fluid One Dimensional Flow Diffusion Elementary theory of diffusion coefficient in gases Water vapour Water vapor and Density of airdry air density calculations at 0 C Ideal gas law Numerical weather prediction Computation Gay Lussac's law Pressure temperature law Hydrostatic equilibrium Astrophysics History of thermodynamics Birth of thermodynamics as science and much more Williams & Meyers Oil and Gas Law Patrick H. Martin, Bruce M. Kramer, 2020 *Williams and Meyers Oil and Gas Law* Patrick H. Martin, Bruce M. Kramer, 2013 *Oil and Gas Law*, 1967 *Oil and Gas Law Review* Christopher B. Strong, 2015 The Oil and Gas Law Review Christopher B. Strong, 2014 *The Oil & Gas Law Review*, 2016 **Oil and Gas Law in a Nutshell** John S. Lowe, 2003 *Cases and Materials on Oil and Gas Law* John S. Lowe, Owen L. Anderson, Ernest Edgar Smith, David E. Pierce, Christopher Stewart Kulander, 2013 *The Oil and Gas Law Review* Ferdinand Adadzi, 2020

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