

Determination of the Solubility-Product Constant for a Sparingly Soluble Salt

A. Preparation of a Calibration Curve

Initial $[\text{CrO}_4^{2-}]$ 0.0024M

Volume of 0.0024 M K_2CrO_4	Total volume	$[\text{CrO}_4^{2-}]$	Absorbance
1. <u>1 mL</u>	<u>50 mL</u>	<u>0.00048M</u>	<u>0.190</u>
2. <u>5 mL</u>	<u>50 mL</u>	<u>0.0024M</u>	<u>1.039</u>
3. <u>10 mL</u>	<u>50 mL</u>	<u>0.0048M</u>	<u>2.127</u>
4. <u>15 mL</u>	<u>50 mL</u>	<u>0.0072M</u>	<u>2.409</u>

Molar absorption coefficient for $[\text{CrO}_4^{2-}]$

1. _____ 2. _____ 3. _____ 4. _____

Average molar absorption coefficient _____

Standard deviation (show calculations) _____

B. Determination of the Solubility-Product Constant

Absorbance	$[\text{CrO}_4^{2-}]$	$[\text{Ag}^+]$	K_{sp} of Ag_2CrO_4
1. <u>0.603</u>	_____	_____	_____
2. <u>0.753</u>	_____	_____	_____
3. <u>1.286</u>	_____	_____	_____

Determination Of A Solubility Product Lab Answers

RD Boyd



Determination Of A Solubility Product Lab Answers:

Decoding **Determination Of A Solubility Product Lab Answers**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Determination Of A Solubility Product Lab Answers**," a mesmerizing literary creation penned by a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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