# CRUDE OIL Waxes, Emulsions, and Asphaltenes



J.R. Becker

## Crude Oil Waxes Emulsions And Asphaltenes Paperback November 1 1997

**Gerald Bernard Shea** 

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which are used to produce petroleum based products for numerous applications from industry to medicine from clothing to food industries We can find petroleum based products all around us This book deals with some important topics related to petroleum such as its chemical composition and stability It is well known that the chemical composition of crude oil differs according to the site of production and its grade varies from waxy to asphaltenic crude Both of them are refined to produce different products The stability of crude oil on aging and transportation is governed by several factors and these factors are included within this book Some new technologies for petroleum characterization are also introduced This book is aimed at researchers chemical engineers and people working within the petroleum industry Effect of Inorganic Solids, Wax to Asphaltene Ratio, and Water Cut On the Stability of Water-in-Crude Oil Emulsions Adel M. Elsharkawy, 2006 Emulsions Manar El-Sayed Abdel-Raouf, 2012 Practices and Methods of Preventing and Treating Crude-oil Emulsions Gerald Bernard Shea, 1939 Crude Oil Emulsion Stability Determination and Its Implications in Enhanced Oil Recovery Xiuyu Wang, 2010 Water in crude oil emulsions can form during oil production especially when enhanced oil recovery EOR methods are applied including Alkaline Surfactant AS Alkaline Surfactant Polymer ASP flooding and possibly LoSal TM water flooding The pros and cons of emulsion formation have motivated significant research in the oil industry. The ultimate objective of this study is to better understand water in crude oil emulsions stabilizing mechanisms. This work is a step further towards engineering and designing conditions that promote recovery based on emulsions for commercially applied and proposed EOR methods I carried out water in oil emulsion stability analysis based on bottle tests at controlled centrifugal force and temperature and an electrorheological technique developed in this project Several parameters affecting the stability of water in oil emulsions were examined including oil to water ratio oil type and rheology aqueous phase composition and temperature By changing oil to water ratio we found that the higher the oil fraction used in preparing the emulsions the more stable the emulsions were The stability comparison for two crude oils with different asphaltene content showed that the heavier oil with higher asphaltene content produced significantly more stable emulsions for all brines tested Brine composition for EOR formulations often has to be tuned to allow for the design of stable formulations in chemical flooding Our results showed that the lower the brine ionic strength the higher the stability of the crude oil emulsions The effect of cation species in the brine was examined by comparing the stability of emulsions made with either purely monovalent or divalent salts to yield the same overall ionic strength It is observed that divalent cations in the brine contributed to the formation of more stable emulsions than monovalent cations for all oils tested The stability of water in oil emulsions is controlled by either the oil film rheology between approaching drops which is mainly a function of the continuous phase viscosity or and by the strength of drop interfacial surfaces In our work temperature is regulated to control the viscosity of the continuous phase and hence determine its effect on emulsion stability through film drainage in contrast with interfacial strength control Additionally stabilization mechanisms for water in crude oil emulsions were investigated

Given the molecular complexity of crude oils more than one component is likely responsible for emulsion stability and we speculate that stability is the outcome of competing material adsorption on water crude oil interfaces For two oils examined the presence of soluble acids and their complexes could be the dominant emulsion stabilizing agents while for the heaviest oil used asphaltenes are possibly more active than organic acids in crude oil to stabilize emulsions Our results suggest that divalent cations may form more active surface complexes with naphthenic acid and or asphaltenes than monovalent cations

<u>Characterization of Asphaltenes and Resins Separated from Water-in-Crude Oil Emulsions Formed in Kuwaiti Oil Firelds</u>
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