

Diagnostische toets

H9 Redoxchemie

Deze toets bestaat uit 13 vragen.

Bij deze toets kun je maximaal 28 punten behalen.

Voor elk vraagnummer zijn de te behalen punten aangegeven.

Kerncentrale

Uraan(IV)oxide, $\text{UO}_2(\text{s})$, wordt gewonnen uit uraanerts voor gebruik in kerncentrales.

Uit uraanerts wordt eerst een ander uraanoxide verkregen: $\text{U}_3\text{O}_8(\text{s})$. In dit uraanoxide zijn uraanionen met verschillende elektrovalenties aanwezig. $\text{U}_3\text{O}_8(\text{s})$ wordt met behulp van een redoxreactie omgezet in uraan(IV)oxide, $\text{UO}_2(\text{s})$.

- 3p **1** Stel zelf de halfreactie op van de omzetting van $\text{U}_3\text{O}_8(\text{s})$ in $\text{UO}_2(\text{s})$. In de halfreactie komen naast de genoemde stoffen ook H_2O en H^+ voor.
- 2p **2** Leg uit of voor de volledige omzetting van uraanoxide, $\text{U}_3\text{O}_8(\text{s})$, in uraan(IV)oxide, $\text{UO}_2(\text{s})$, een oxidator of reductor nodig is.

Zink

Zink is geen edelmetaal, maar toch hoeven zinken voorwerpen niet te worden beschermd tegen corrosie.

- 2p **3** Geef hiervoor een verklaring.

Zink kan op verschillende manieren als bescherm laag worden aangebracht. Een voorwerp kan thermisch of elektrolytisch worden verzinkt. Het voordeel van elektrolytisch verzinken is dat het een zeer dun en glad laagje kan aanbrengen. Bij een bedrijf kan men staal voorzien van een laagje zink van slechts enkele micrometers dik. Hiervoor wordt het stalen voorwerp twaalf uur lang elektrolytisch verzinkt in een zinksulfaatoplossing met een stroomsterkte van 2,5 A.

- 4p **4** Bereken hoeveel gram zink hierbij wordt verbruikt.

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TD Snyder

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'TD Snyder' text.

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Table of Contents Diagnostische Toets Redoxreacties

1. Understanding the eBook Diagnostische Toets Redoxreacties
 - The Rise of Digital Reading Diagnostische Toets Redoxreacties
 - Advantages of eBooks Over Traditional Books
2. Identifying Diagnostische Toets Redoxreacties
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Diagnostische Toets Redoxreacties
 - User-Friendly Interface
4. Exploring eBook Recommendations from Diagnostische Toets Redoxreacties
 - Personalized Recommendations
 - Diagnostische Toets Redoxreacties User Reviews and Ratings
 - Diagnostische Toets Redoxreacties and Bestseller Lists
5. Accessing Diagnostische Toets Redoxreacties Free and Paid eBooks
 - Diagnostische Toets Redoxreacties Public Domain eBooks
 - Diagnostische Toets Redoxreacties eBook Subscription Services
 - Diagnostische Toets Redoxreacties Budget-Friendly Options

6. Navigating Diagnostische Toets Redoxreacties eBook Formats
 - ePub, PDF, MOBI, and More
 - Diagnostische Toets Redoxreacties Compatibility with Devices
 - Diagnostische Toets Redoxreacties Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Diagnostische Toets Redoxreacties
 - Highlighting and Note-Taking Diagnostische Toets Redoxreacties
 - Interactive Elements Diagnostische Toets Redoxreacties
8. Staying Engaged with Diagnostische Toets Redoxreacties
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Diagnostische Toets Redoxreacties
9. Balancing eBooks and Physical Books Diagnostische Toets Redoxreacties
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Diagnostische Toets Redoxreacties
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Diagnostische Toets Redoxreacties
 - Setting Reading Goals Diagnostische Toets Redoxreacties
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Diagnostische Toets Redoxreacties
 - Fact-Checking eBook Content of Diagnostische Toets Redoxreacties
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

-
- Interactive and Gamified eBooks

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