

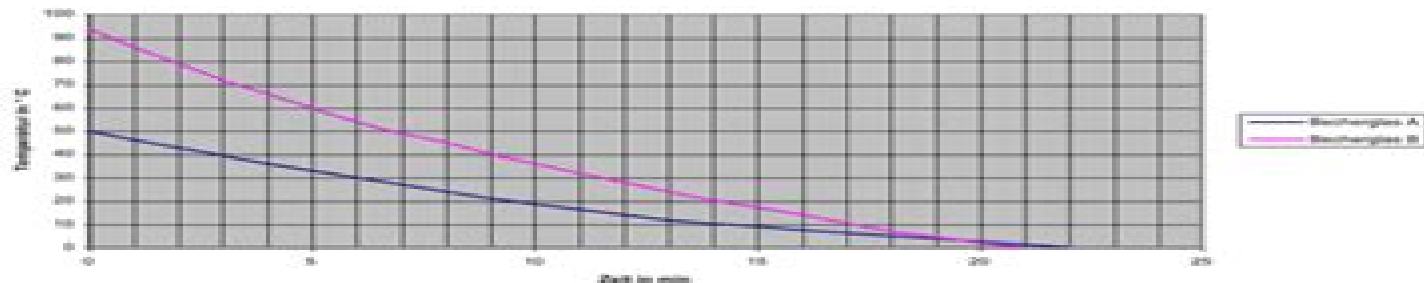
Verdampfungsprozess – Versuch Gefrieren von Wasser

Praktische Anmerkungen zum Versuch

In der Versuchsanleitung wurde eine Kühlmischung mit zerhacktem Eis und Salz im Verhältnis 3 zu 1 hergestellt. Das ist nur eine Variante der Durchführung. Es kann ein durchsichtiger Kühlschrank gebraucht werden. Auch wurde dieser Versuch in einer oben offenen Kühltruhe erfolgreich durchgeführt. Dabei kann man den Deckel leicht öffnen und so zu den Thermometern im Innern blicken. Nicht eignen sich Kühlschränke, da die Kälte sofort nach unten wegfließt. Eine weitere Möglichkeit eröffnen Kälte Tage oder Nächte, man kann diesen Versuch problemlos im Freien bei höchstens $-5\text{ }^{\circ}\text{C}$ durchführen. Die Daten können auch elektronisch erfasst werden, und wenn das Gerät es erlaubt, kann der Taschenrechner mit Interface in den geschlossenen Kühlschrank gestellt werden.

Theoretische Anmerkungen zum Versuch

Die Abkühlung beider Bechergläser ist nicht linear. Dies wird nur zur Veranschaulichung für die Schüler so beschrieben. Beim Durchführen entstand folgende Kurve:



Am Anfang der Kurve entspricht die Kurve den Erwartungen, also je wärmer desto schnelleres Sinken der Temperatur. Doch in diesem Falle müssten die Kurven zumindest nach deren Schnittpunkt identisch sein, denn beide Flüssigkeiten sind gleich warm. Nun ist zu erkennen, dass die Menge des verdampften Wassers so unterschiedlich ist, dass sich die kleinere Menge ganz einfach schneller kühlen lässt. Sehen wir den Punkt 50°C bei beiden Kurven an. Die Kurve des Becherglas B ist darin viel steiler fallend, d.h. schneller abkühlend, als die Kurve des Becherglases A. Also muss schon hier der Masseunterschied beträchtlich sein. Als Beleg wurden die Bechergläser gewogen:

	Nach Erwärmungsphase:		Nach Abkühlungsphase:		Verlust
Ausgangslage	20°C	100 g			
Becherglas A	50°C	97 g	0°C	90 g	10 %
Becherglas B	99°C	91 g	0°C	75 g	25 %

Diese Zahlen sind stark abhängig von der Dauer des Experimentes, so sind die Verlustwerte bei schnellem Abkühlen viel tiefer.

Es ist also der Massenverlust und das Abkühlen beim Verdampfen, dass dazu führt dass das Wasser in Becherglas B zuerst gefriert.

Das Lexikon K Mpfens Erkl Rungen Selbstverteidigung

CL Gary



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