

Source: *Author's calculations*.

10. What does it mean that the  $\text{C}_{60}$  molecule is a truncated icosahedron? (Answer: see attached sheet)
11. How many atoms are there in  $\text{C}_{60}$ ?
12. How many  $\text{C}-\text{C}$  bonds are there in  $\text{C}_{60}$ ?
13. How many  $\text{C}-\text{C}$  bonds are there in  $\text{C}_{60}$  that are  $\text{C}-\text{C}$  single bonds?
14. How many  $\text{C}-\text{C}$  bonds are there in  $\text{C}_{60}$  that are  $\text{C}=\text{C}$  double bonds?
15. How many  $\text{C}-\text{C}$  bonds are there in  $\text{C}_{60}$  that are  $\text{C}-\text{C}$  single bonds?
16. How many  $\text{C}-\text{C}$  bonds are there in  $\text{C}_{60}$  that are  $\text{C}=\text{C}$  double bonds?
17. How many  $\text{C}-\text{C}$  bonds are there in  $\text{C}_{60}$  that are  $\text{C}-\text{C}$  single bonds?
18. How many  $\text{C}-\text{C}$  bonds are there in  $\text{C}_{60}$  that are  $\text{C}=\text{C}$  double bonds?
19. How many  $\text{C}-\text{C}$  bonds are there in  $\text{C}_{60}$  that are  $\text{C}-\text{C}$  single bonds?
20. How many  $\text{C}-\text{C}$  bonds are there in  $\text{C}_{60}$  that are  $\text{C}=\text{C}$  double bonds?

# Crucible Final Exam Study Guide Answers

**R Sandford**



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