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# Balancing an equation that contains a polyatomic ion example problem

$$Al_{(n)} + NaNO_{3(nq)} \rightarrow Al(NO_3)_{3(nq)} + Na_{(n)}$$

Step 1: Since there is an unequal number of N and O on both sides, the equation is unbalanced.

Tay NO, is a polyatomic ion - when balancing keep the "NO, together as a unit for balancing."

Step 2: Working from left to right add a "3" in front of NaNO, to balance NO,) and a "3" in front of the Na in the products.

Step 3: Recheck - The equation is balanced!

Almi + SNaNO com -> Al(NO) library + SNapo

# Balancing Organic Combustion reactions example problem

C.H.JE + D. + COURT + H.OSE

Sings to Phase and in November COS, saled a 3 in-home of PLCs. This year bot

C.H., ID + OL + 400 (MD + 744,000)

Shop 2: Simps an indictestal exember of pages atoms is present on the greature side (FQ + 7) = 19 angree atoms. A "2" example to placed in fruint of C\_PL, and the present on CQ, and P<sub>Q</sub>O is changed to 12 and 14. This section

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# Balancing an equation that contains a polyatomic ion try it problems

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Jianjun Gao

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