11.1 - Describing Chemical Reactions

- 1. What two substances reacted in the Hindenburg crash?
- 2. In general, what happens in a chemical reaction?
- Look at Figure 11.1. What are some of the evidences of chemical change (a.k.a. chemical reaction)?
- Give an example of a word equation for a chemical reaction (you can make one up or choose one from the book).
- 5. How is a chemical equation different from a word equation?
- 6. What is missing in a skeleton equation?
- 7. What symbols are used to represent the phases of substances in chemical reactions?
- 8. What does the symbol (ag) mean?
- 9. Use the space below to do Practice Problems #1 and #2.

 1)
- A complete description of a reaction must include

but also _____

- 11. Define the following:
 - a. coefficients -
 - h. subscripts -
 - c. balanced equation -
- 12. How does the law of conservation of mass relate to belancing reaction equations?
- 13. In a chemical reaction, what MUST be the same for the reactants and the products? (There are two great answers to this question!)

Chapter 11 Guided Notes Name 11 1 A Describing Chemical

Rachel Sandford

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