

Name _____ Date _____ Class _____

CHAPTER 15 **STUDY GUIDE FOR CONTENT MASTERY**

Solutions

Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

Use each of the terms below just once to complete the passage.

immiscible	liquid	soluble	solution
insoluble	miscible	solvent	solvent

Air is air! (1) solution of oxygen gas dissolved in nitrogen gas. The oxygen is air is the (2) solvent and nitrogen is the (3) solvent. Because oxygen gas dissolves in a solvent, oxygen gas is air is (4) soluble substances. A substance that does not dissolve is (5) insoluble. (6) liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are (7) miscible. However, if one liquid is insoluble in another, the liquids are (8) immiscible.

Read about solvation in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



8. Hydration is solvation in which the solvent is water. What is solvation?

Solvation is the process of surrounding solute particles with solvent particles to form a solution.

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CHAPTER 15 **STUDY GUIDE FOR CONTENT MASTERY**

Section 15.1 continued

10. As sodium chloride dissolves in water, what happens to the sodium and chloride ions?

The sodium and chloride ions are separated and surrounded by the water molecules.

11. Explain the attraction of the water molecules around the sodium ions and chloride ions.

Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

12. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solvation process to occur.

13. List three ways that the rate of solvation may be increased.

stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write true or false.

true 14. The overall energy change that occurs when a solution forms is called the heat of solution.

false 15. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.

false 16. Solvation continues as long as the solvation rate is less than the crystallization rate.

true 17. In a saturated solution, solvation and crystallization are in equilibrium.

true 18. Additional solute can be dissolved in an unsaturated solution.

true 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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Chapter 15 Study Guide For Content Mastery Solutions Answers

M. Linda Workman, Linda A. LaCharity



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Understanding Pharmacology - E-Book M. Linda Workman, Linda A. LaCharity, 2015-10-29 Pharmacology can be difficult But with the right text understanding drugs and how they work doesn't have to be Using easy to follow language and engaging learning tools like Memory Joggers Clinical Pitfalls Do Not Confuse and Drug Alerts the second edition of Understanding Pharmacology Essentials for Medication Safety helps readers really understand how drugs work In addition to the popular critical thinking activities from the first edition the second edition also includes more chapter review questions updated content and a new organization that centers on the different body systems For students who have a limited background in the sciences and want complete preparation for licensure exams and clinical practice there is no better choice than Understanding Pharmacology 2nd Edition Entire unit reviewing math weights and measures and dosage calculation minimizes readers anxiety and promotes medication safety Clever easy to recognize margin icons help visual learners remember essential side effects of drugs Simplified heading structure replaces intimidating terminology i e pharmacokinetics with simplified language How These Drugs Work to increase understanding of concepts Drug Alert Do Not Confuse and Clinical Pitfall boxes highlight important tips for safe medication administration Memory Jogger boxes help readers remember important drug information Get Ready for Practice sections at the end of each chapter include key points chapter review questions and critical thinking activities to reinforce learning 10th grade reading level uses straightforward everyday language to really enhance readers understanding of pharmacology concepts Incorporation of adult learning theory features both a simple to complex organization of material along with answers to why readers need to learn something NEW Body system organization helps readers better understand drugs that are specific to particular body systems NEW More chapter review questions have been added to the text All review questions are now organized into one of two categories Test Yourself on the Basics and Test Yourself on Advanced Concepts

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MSN, CCRN, 2013-01-18 Part of the popular LPN Threads series this market leading text features an easy to follow writing style and organization to teach you the concepts and skills you need to know to practice nursing in a variety of care settings This new edition offers in depth discussion of QSEN competencies rationales for the NCLEX PN review questions and new icons to point out the most current evidence based information Standard LPN Threads features include helpful characteristics such as full color design key terms numbered objectives key points critical thinking questions critical thinking activities glossary and references Easy to follow reading level and text organization presents information from simple to most complex Think Critically boxes help you synthesize information and apply concepts beyond the scope of the chapter Cultural Considerations related to biocultural variations as well as health promotion for specific ethnic groups demonstrate how to provide culturally competent care Elder Care Points highlight changes that occur with aging and how they affect nursing care Home Care boxes highlight the necessary adaptations of nursing skills and techniques for the patient in the home care setting NEW Rationales for NCLEX review questions at the end of each chapter help you understand why your choices were correct or incorrect NEW Full text reviews by experts in the field offer consistency and ease understanding as you progress through the book NEW Evolve margin icons denote supplemental material for students on Evolve NEW Evidence Based Practice margin icons point out the most current and evidence based information NEW In depth discussion of the Quality and Safety Education for Nurses QSEN within the text provides the knowledge skills and attitudes necessary to continuously improve the quality and safety of the healthcare systems

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