

## Chemistry 2 ACS final exam 2024 Comprehensive Review

ideal gas law conditions - **Answer>>** low pressure, high temperature

boyle's law - **Answer>>**  $p_1v_1=p_2v_2$

gases are highly \_\_\_\_\_ - **Answer>>** compressible

gases expand \_\_\_\_\_ to fill the container they are placed in - **Answer>>** spontaneously

the distance between molecules in gases are \_\_\_\_\_ - **Answer>>** far apart in gases

1 mol of any gas at STP is - **Answer>>** 22.4 liters

the lighter the gas, \_\_\_\_\_ - **Answer>>** the faster the effusion rate

the deviation from ideal behavior of a gas is most evident at high \_\_\_\_\_ and low \_\_\_\_\_ - **Answer>>** pressure, temp

Non reacting gas mixtures are homongeneous - **Answer>>**

normal melting point at - **Answer>>** 760 torr or 1 atm

the kinetic-molecular theory predicts that pressure rises as the temperature of a gas increases because \_\_\_\_\_ -

**Answer>>** both the gas molecules collide more frequently with the wall and the gas molecules collide more energetically with the wall

# Chemistry Final Exam Comprehensive Review Guide

**M Mosston**



## **Chemistry Final Exam Comprehensive Review Guide:**

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